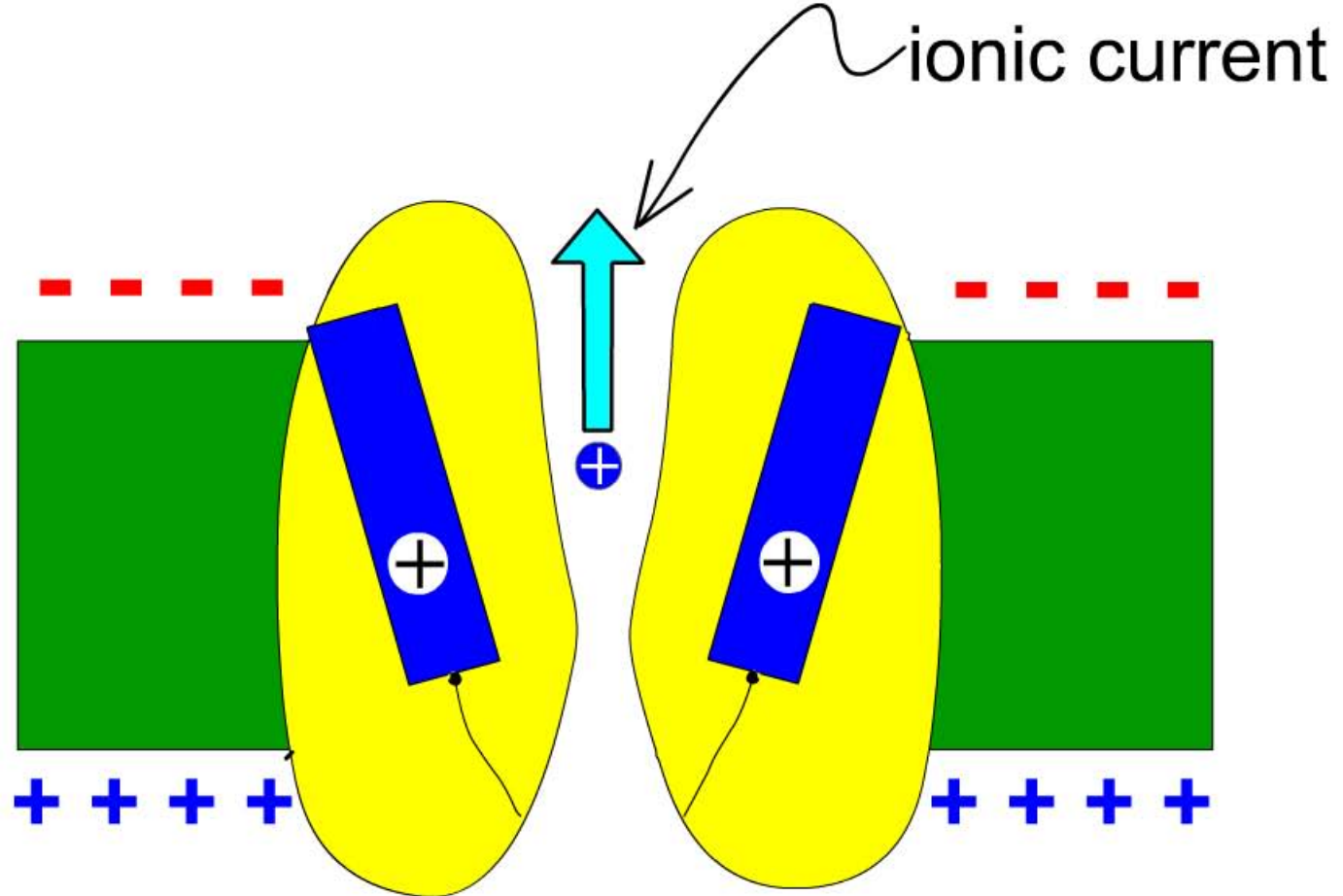


***Learning about
Conformational changes in
ion channels with
Fluorescence***

***Walter Sandtner, Clark Hyde, Jerome
Lacroix, Carlos Villalba-Galea, Baron
Chanda, Rikard Blunck Francisco
Bezanilla***

***Dept. Biochemistry and Molecular
Biology
University of Chicago***

A voltage-gated ion channel



OPEN

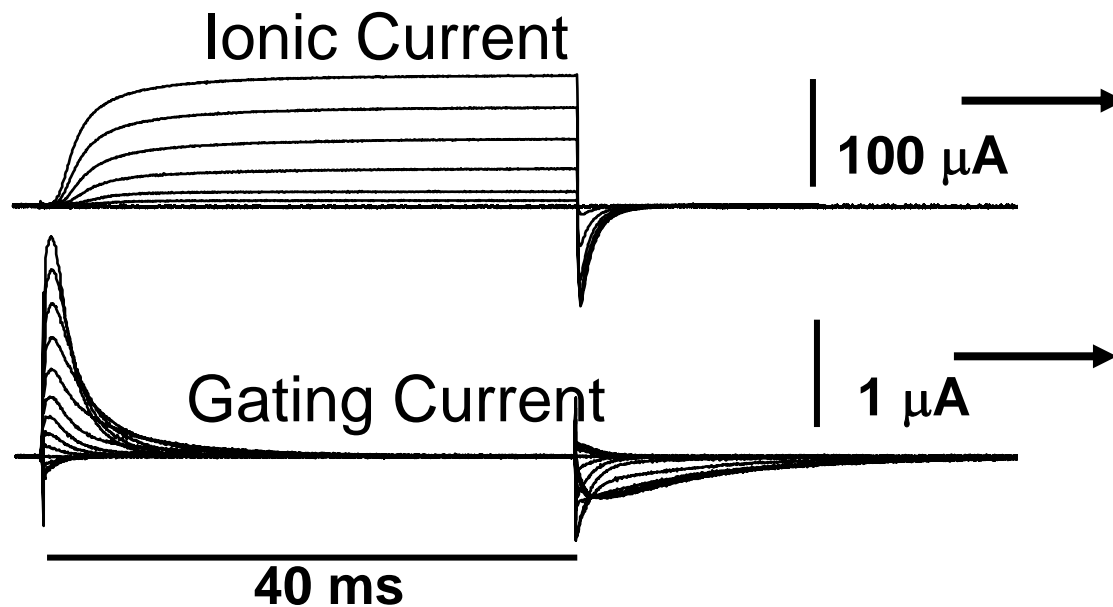
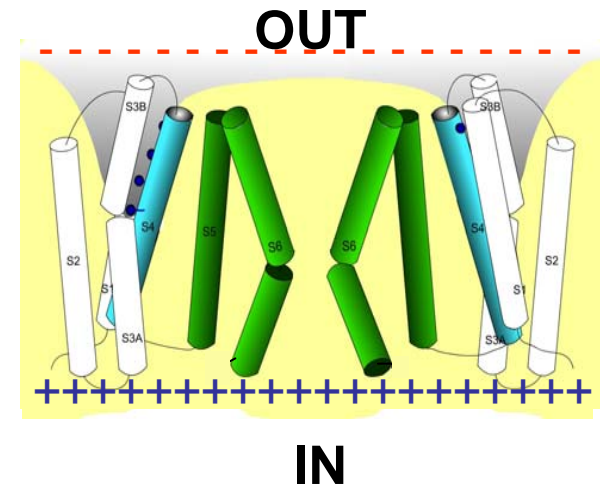


Conformational changes detected at the macroscopic level

Electrophysiological Methods

1. Oocyte Cut-open Voltage clamp
2. Whole cell patch clamp
3. Bilayers voltage clamp
4. Liposomes "Nernst clamp"

What do we record?



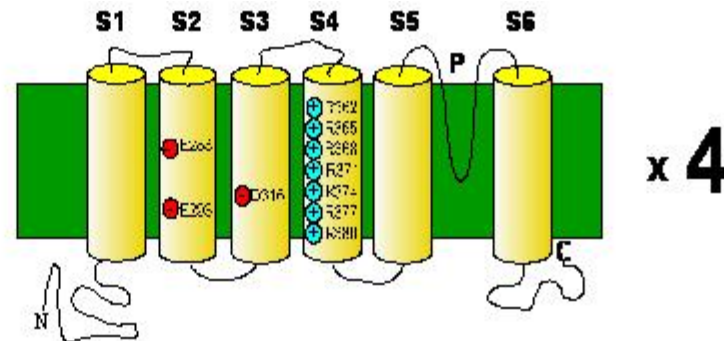
The conduction of ions through the protein:

-----the actual function-----

The rearrangement of intrinsic charges in the protein:

----voltage dependent Conformational changes-----

ShakerB



ShakerB S4 AILR³⁶²VIR³⁶⁵LV³⁶⁸VER³⁷¹IEK³⁷⁴LSR³⁷⁷HSK³⁸⁰

Rat μ 1 DI ISALR²¹⁶TF²¹⁹R²¹⁹VLR²¹⁹AIK²¹⁹ITVI²¹⁹PGLKT

Rat μ 1 DII SVLR⁶⁶⁰SF⁶⁶³RL⁶⁶³LV⁶⁶³FK⁶⁶³LAK⁶⁶³SWPTL

Rat μ 1 DIII LGPIK¹¹¹⁵SL¹¹¹⁹TL¹¹¹⁹R¹¹¹⁹AI¹¹¹⁹R¹¹¹⁹PI¹¹¹⁹R¹¹¹⁹ALSR

Rat μ 1 DIV SPTLF¹⁴³⁶R¹⁴⁴¹VIR¹⁴⁴¹LR¹⁴⁴¹IG¹⁴⁴¹R¹⁴⁴¹VLR¹⁴⁴¹LI¹⁴⁴¹R¹⁴⁴¹GAK¹⁴⁴¹GI¹⁴⁴¹R¹⁴⁴¹

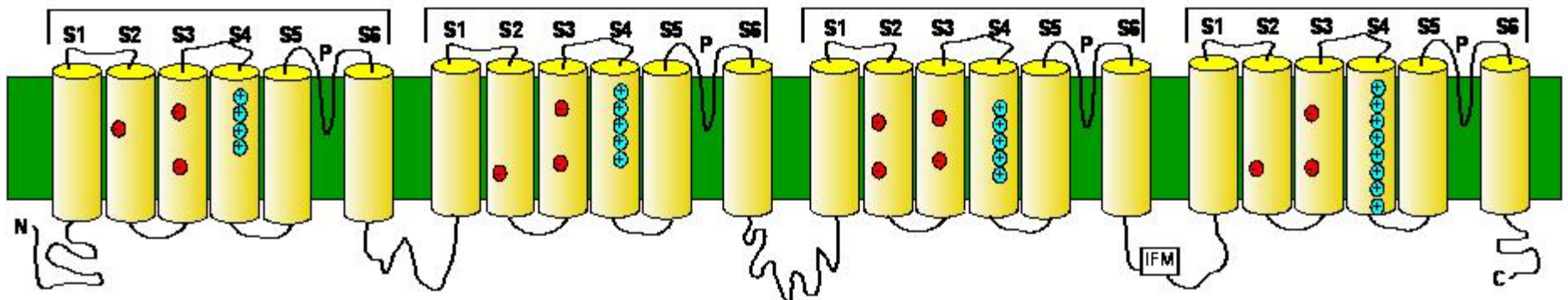
Rat μ 1

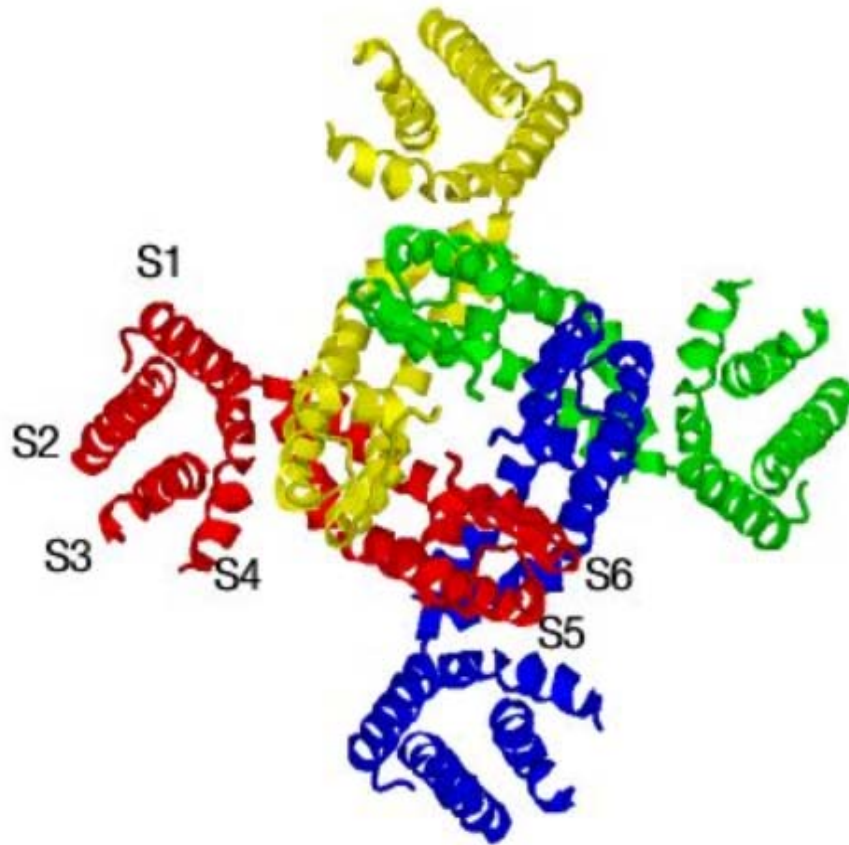
Domain I

Domain II

Domain III

Domain IV

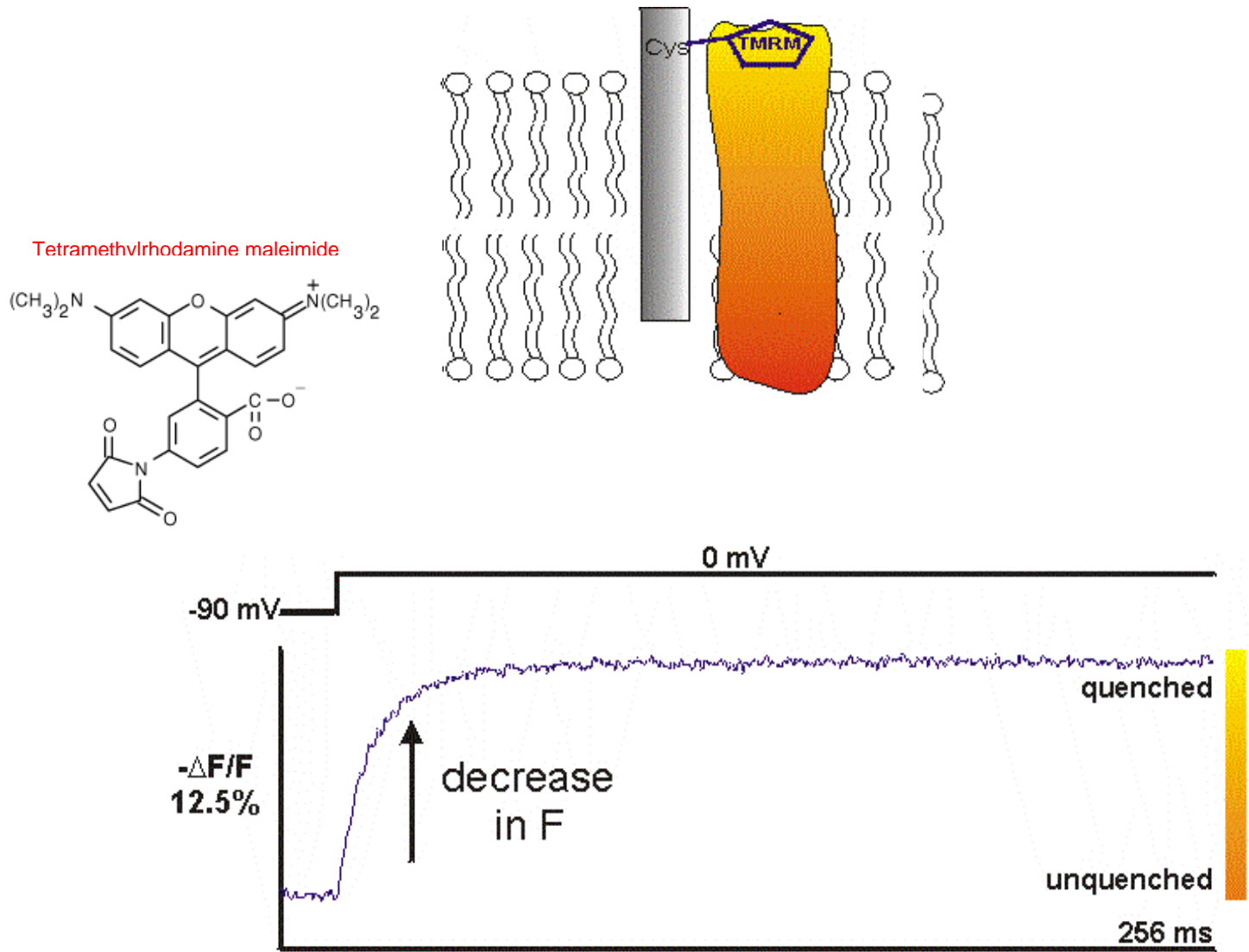


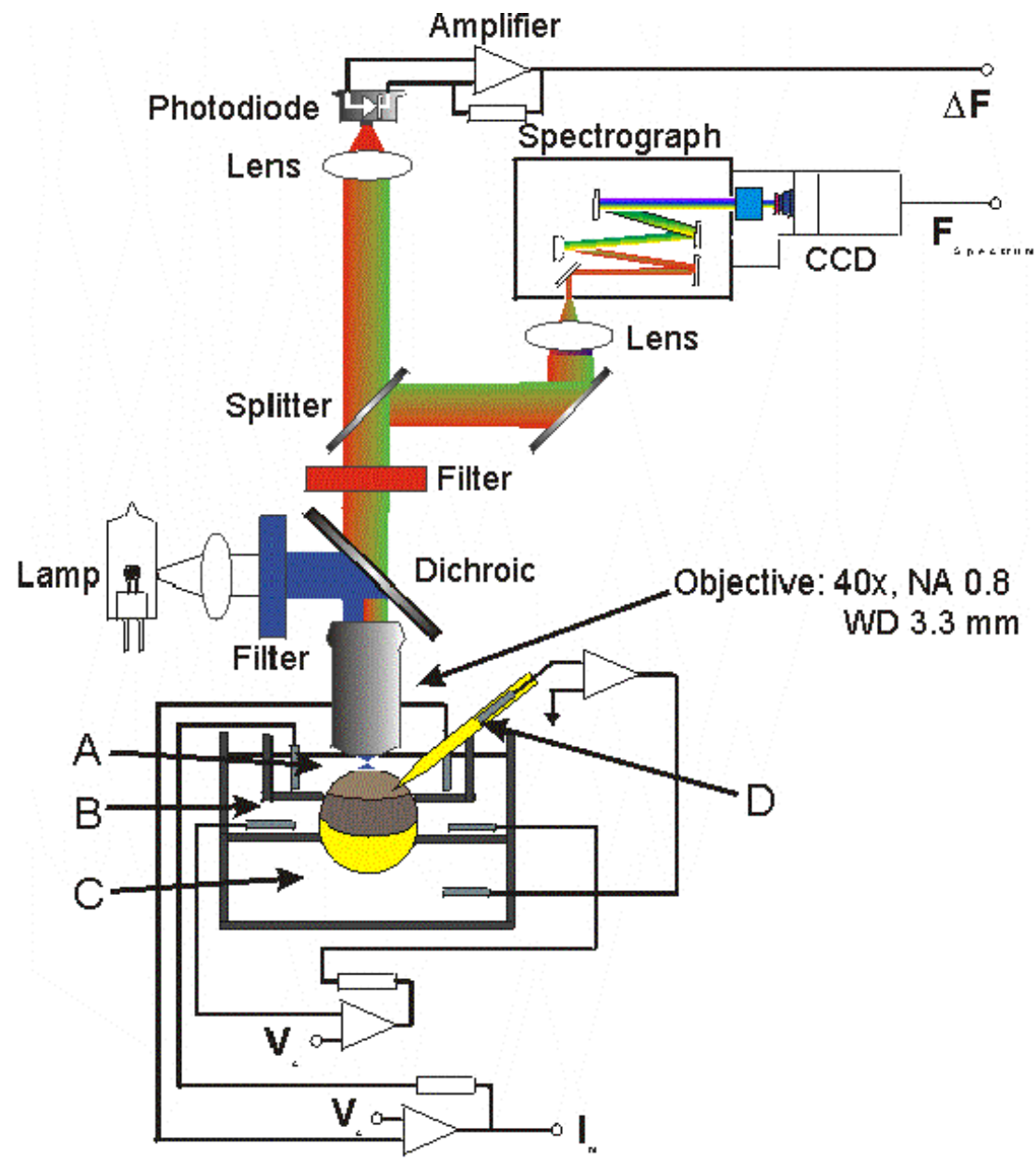


X-ray structure of **Kv1.2** (pdb id 2A79)
Long et al (Science, 2005)

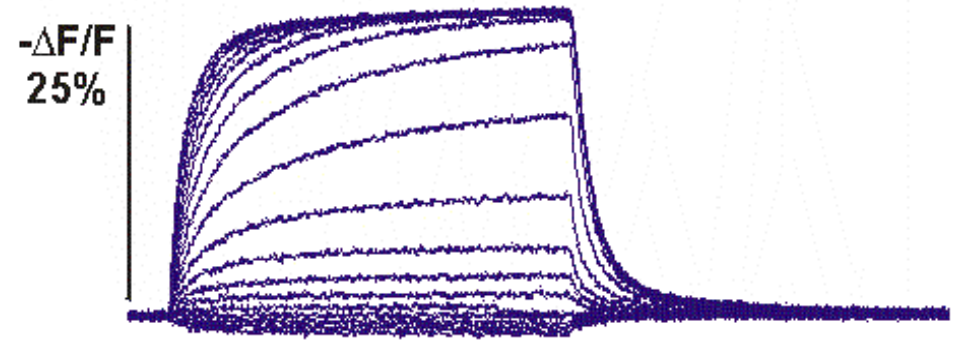
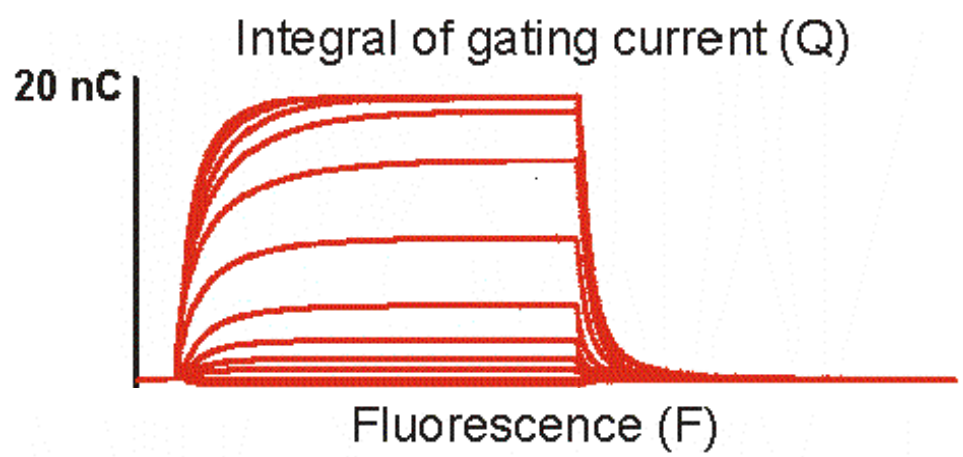
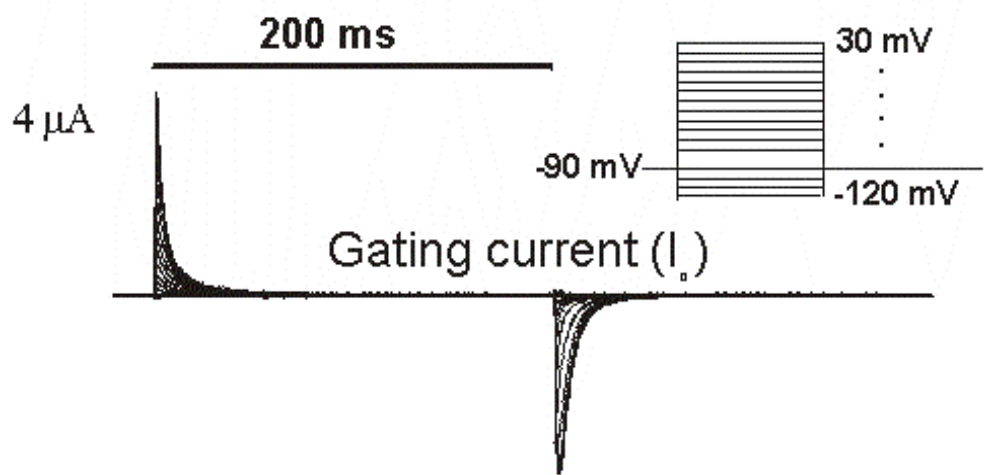
OPEN-INACTIVATED state

Detection of conformational change

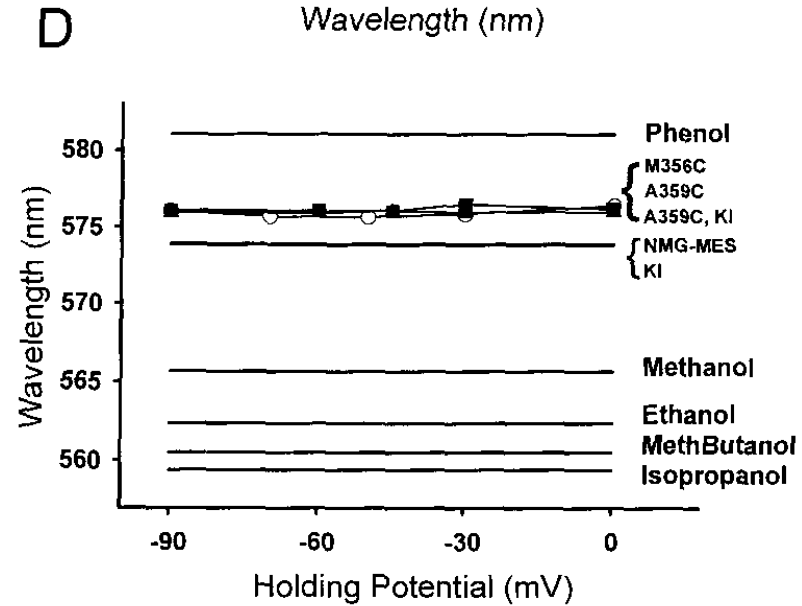
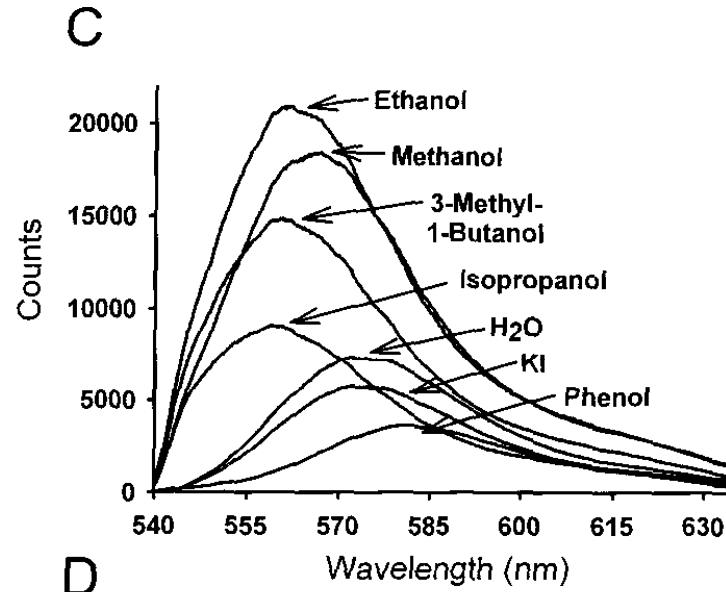
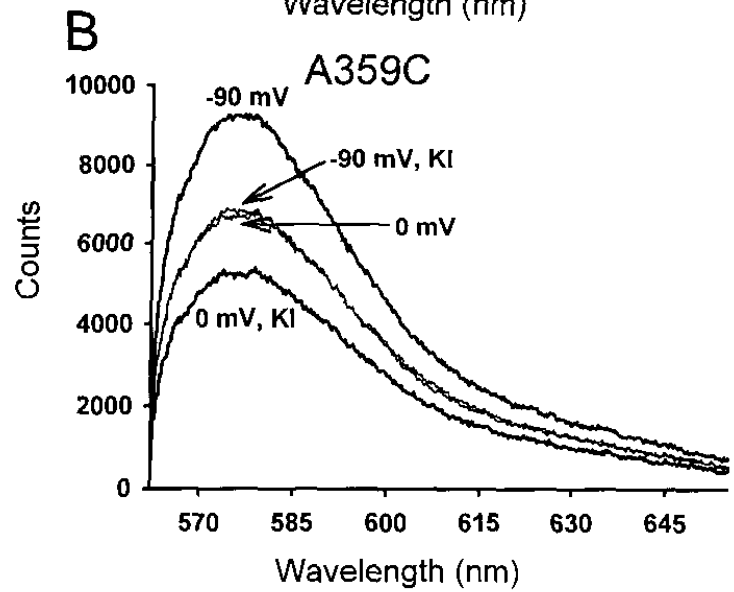
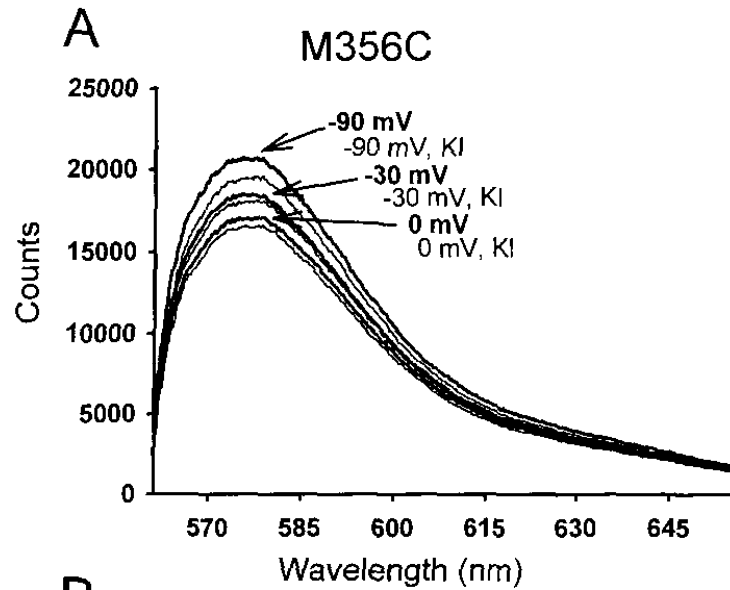




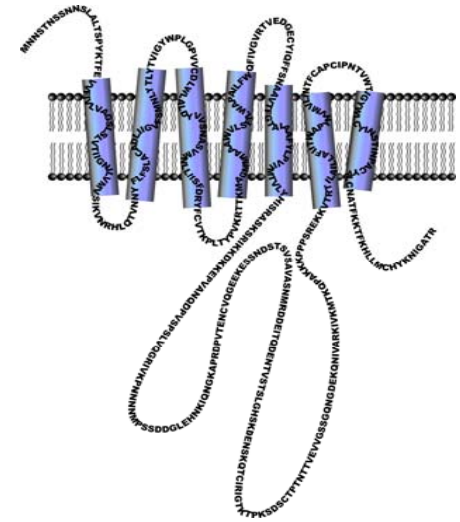
Experimental data: M356C W434F



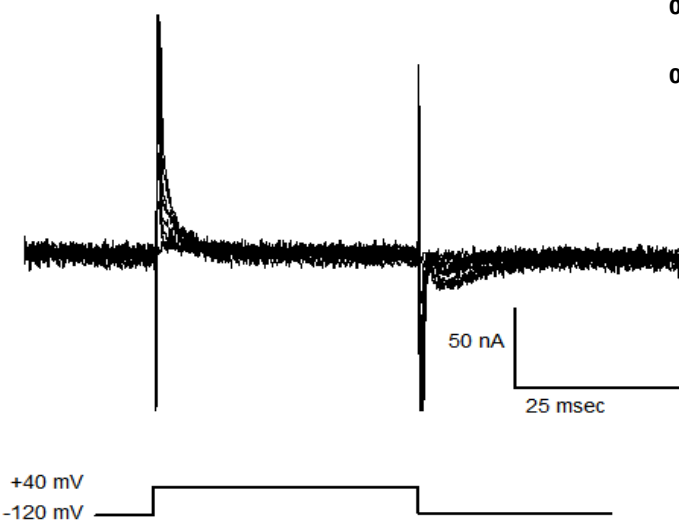
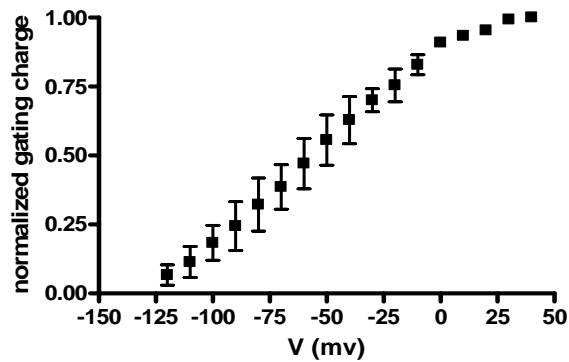
The fluorescence change in M356C is due to quenching



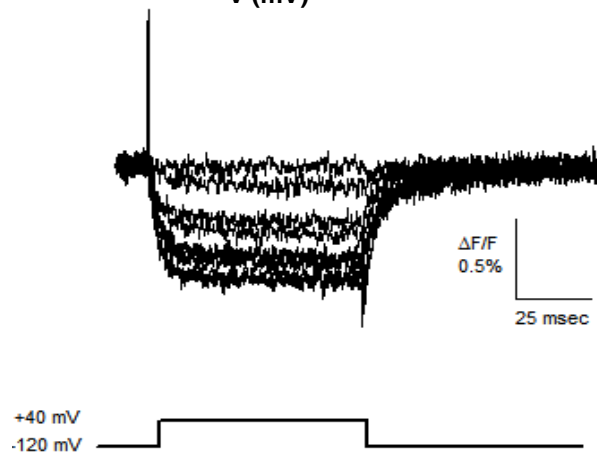
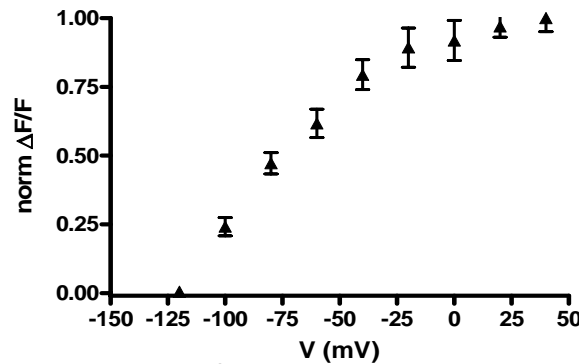
Depolarization Induces a Conformational Change in the m2 Muscarinic Receptor



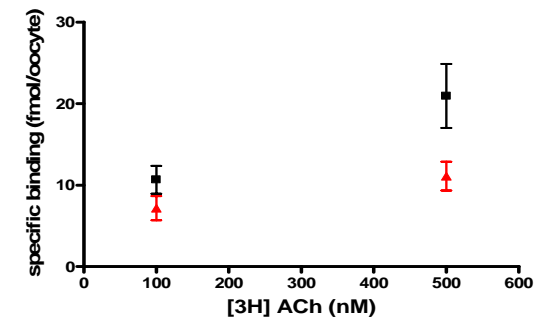
Gating currents



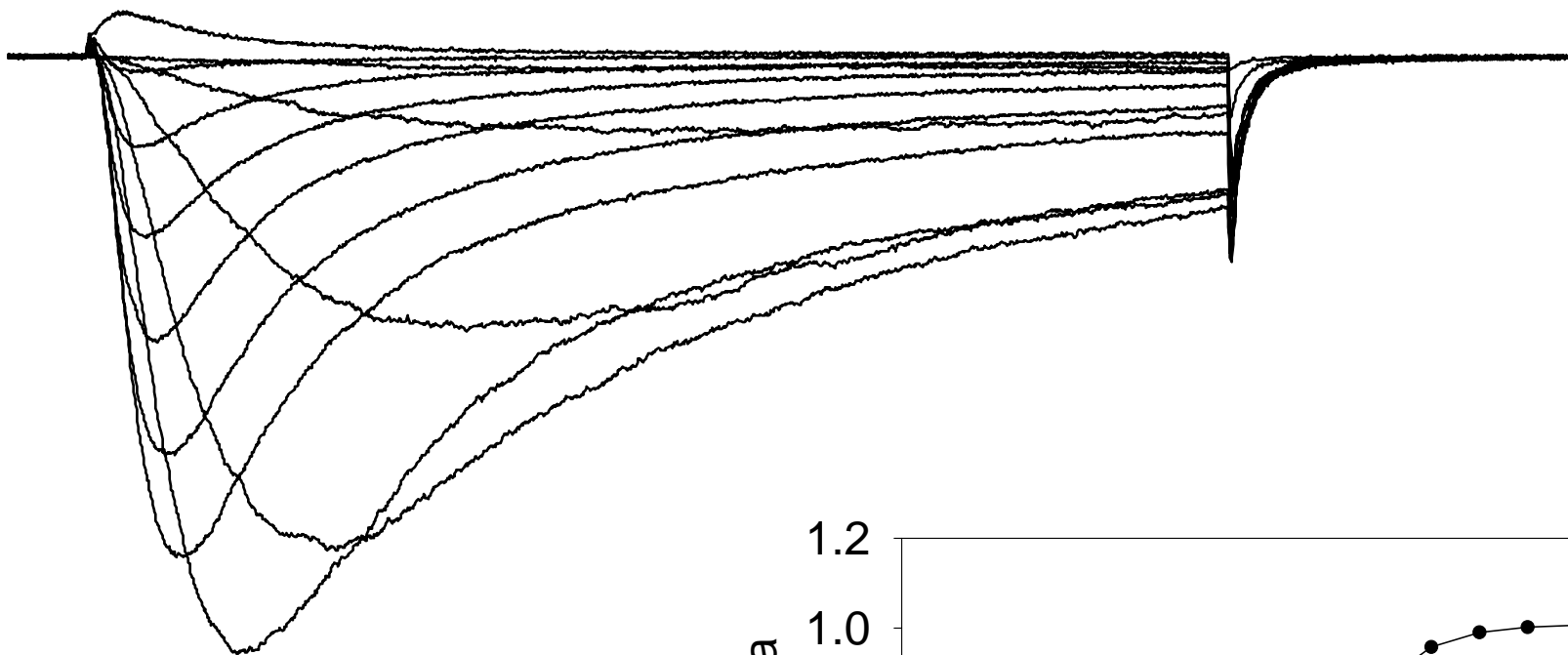
Fluorescence signal



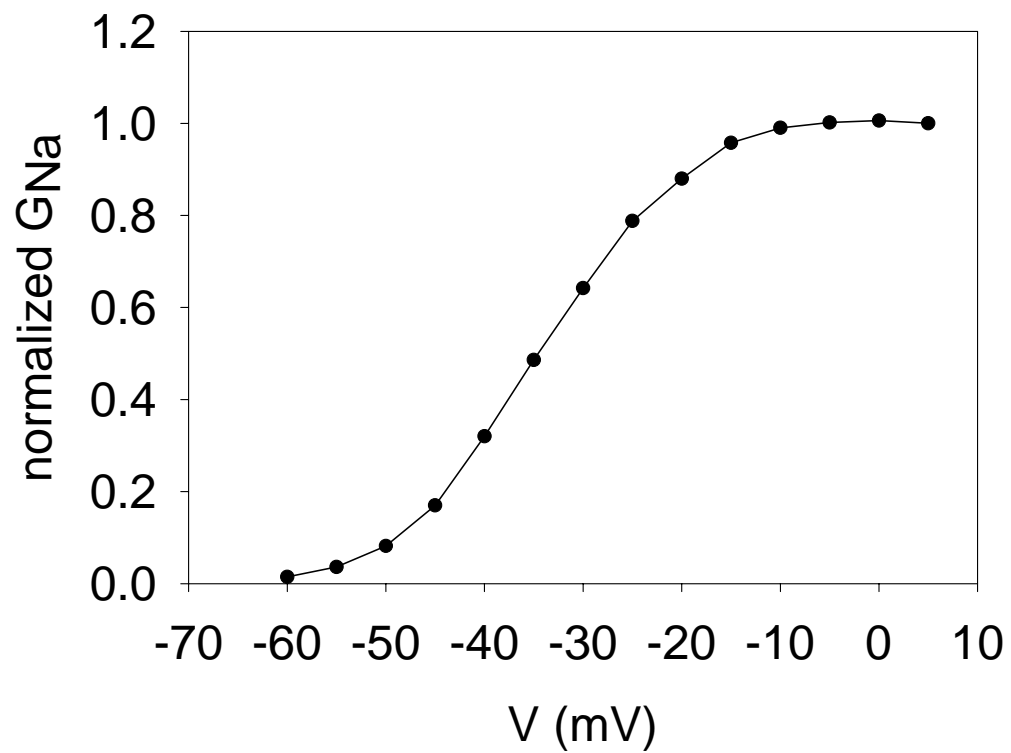
Voltage dependent binding to *Xenopus* oocytes



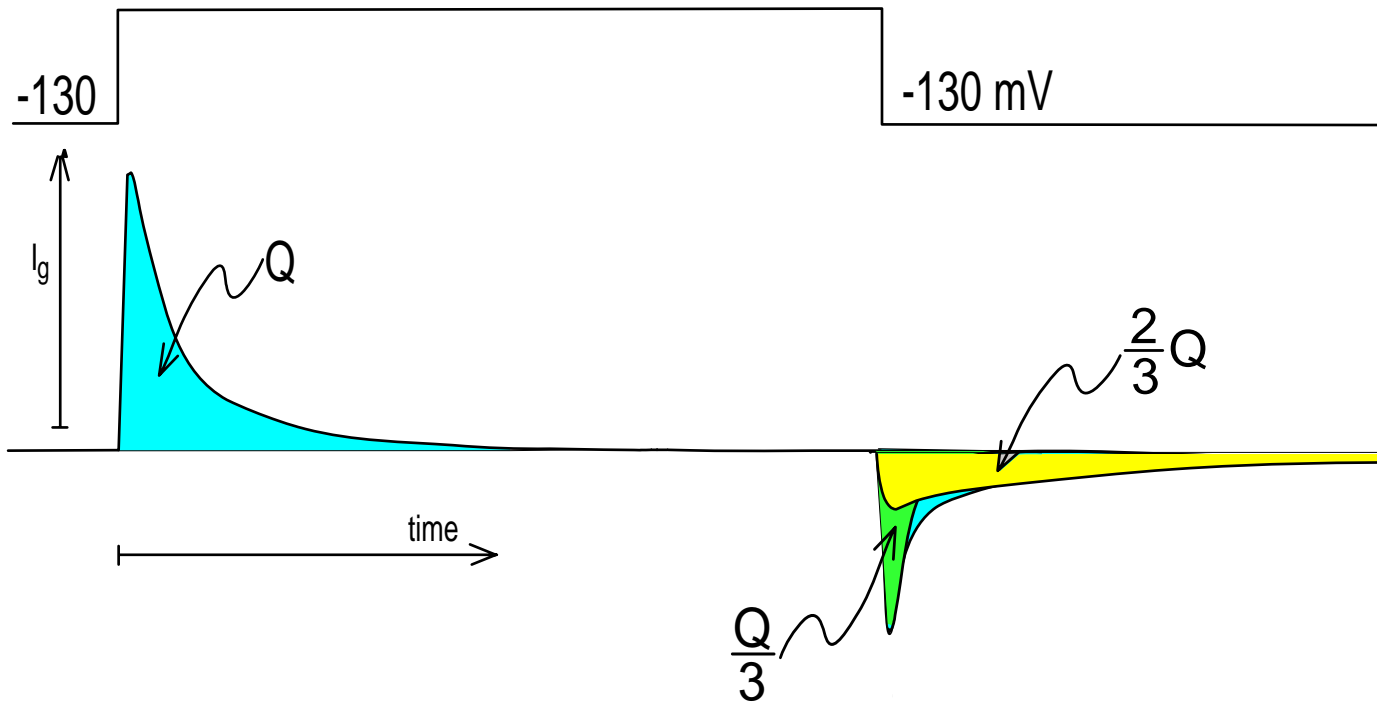
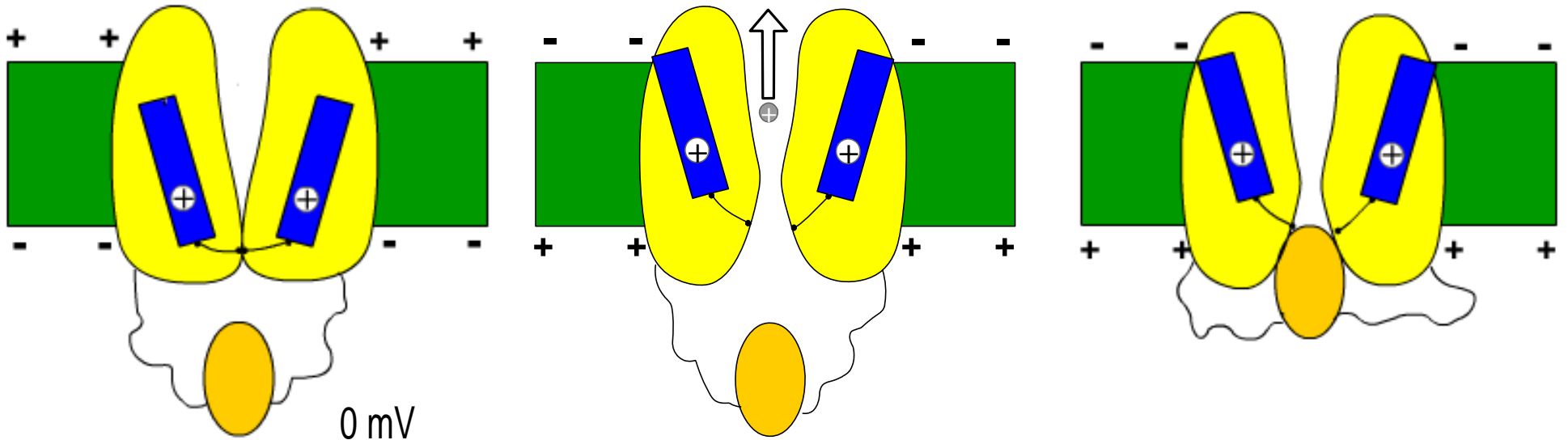
- resting potential
- ▲ depolarization

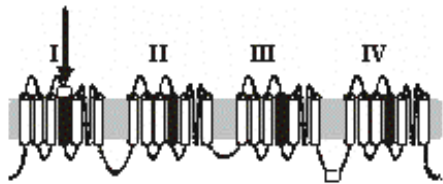


Na Channel

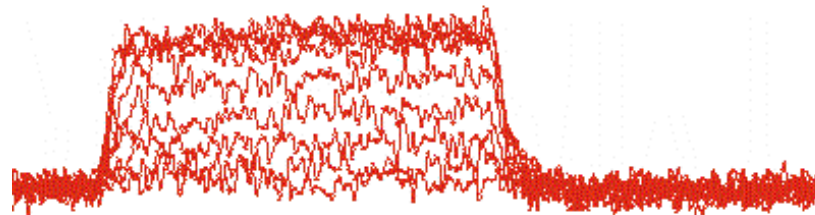
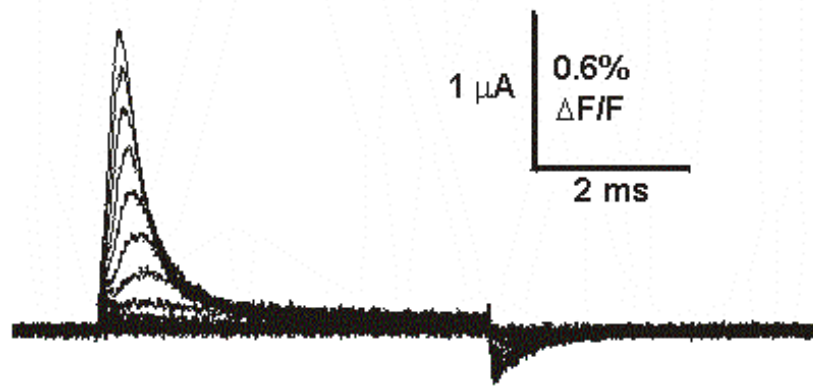


Sodium Channel

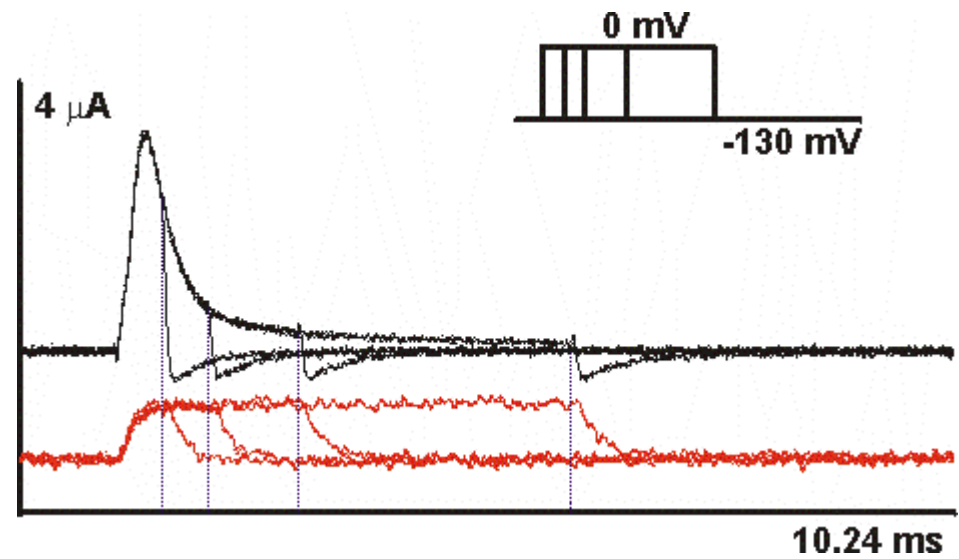
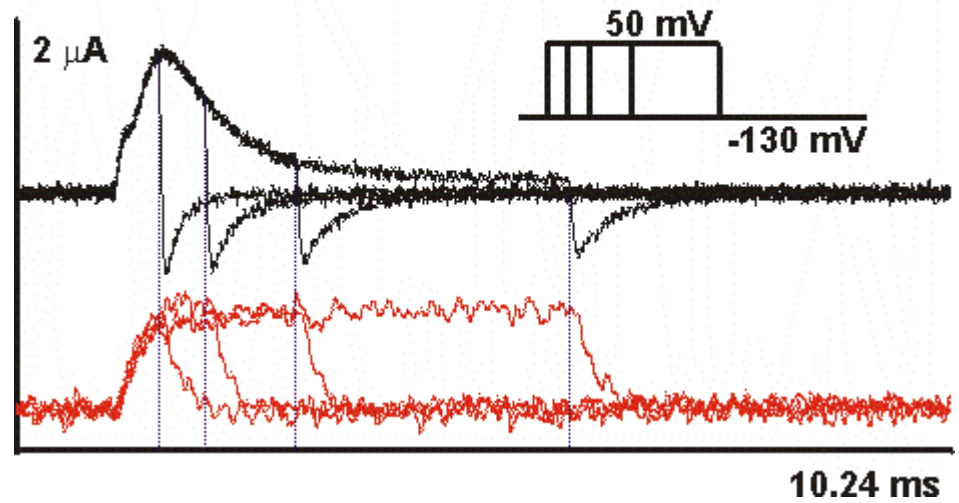




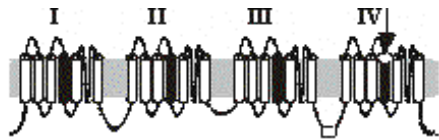
Domain I (S216C)



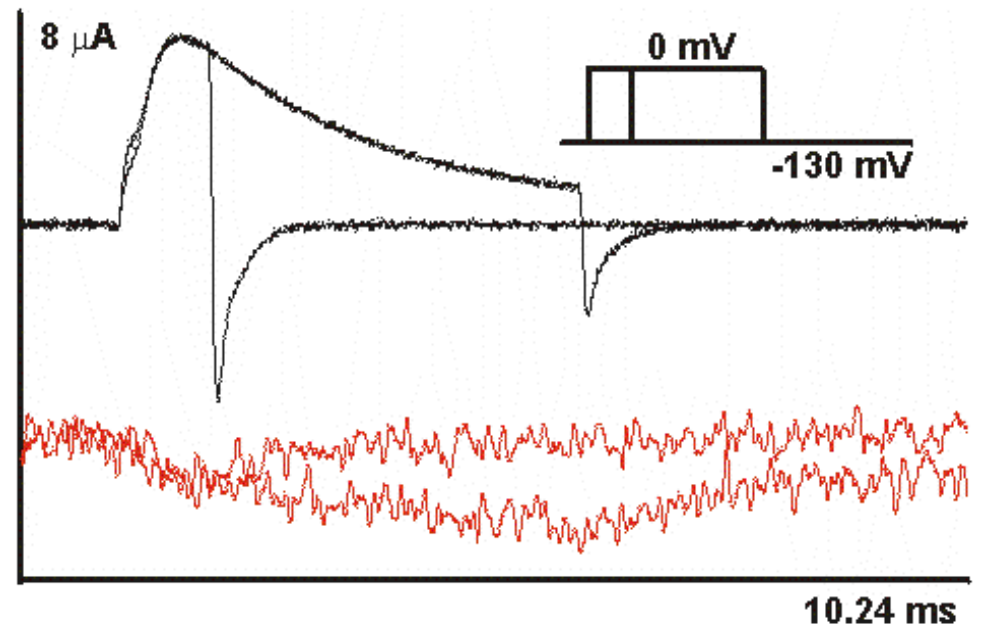
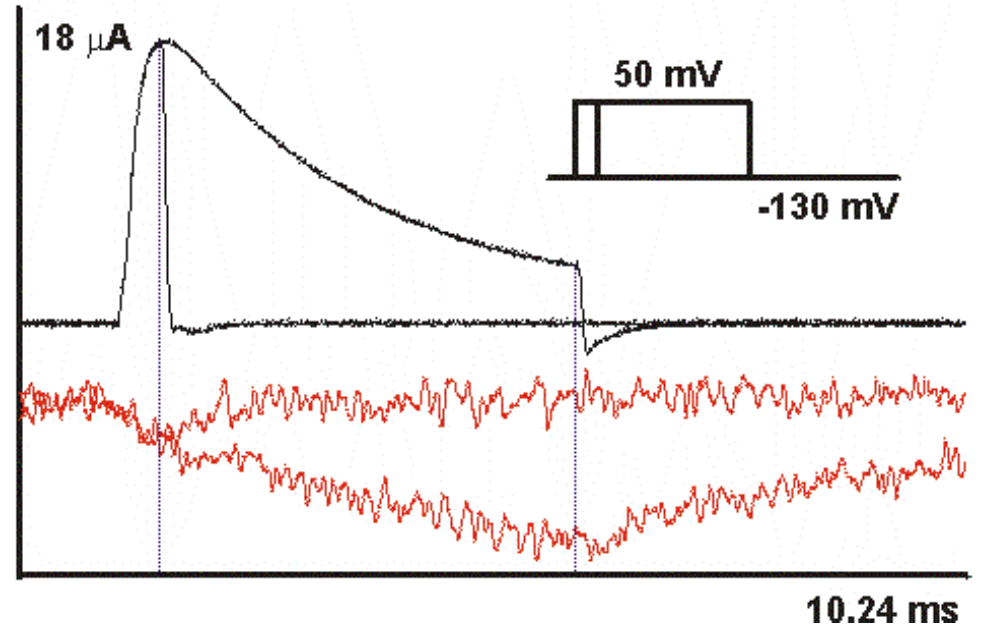
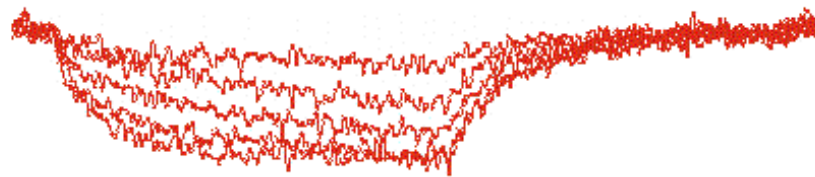
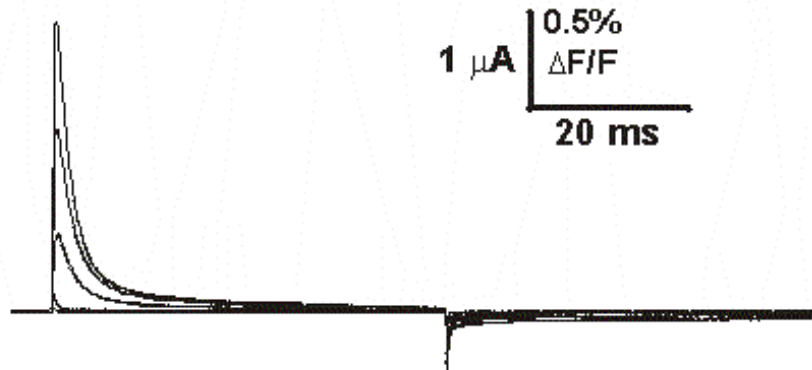
Lack of immobilization in domain I

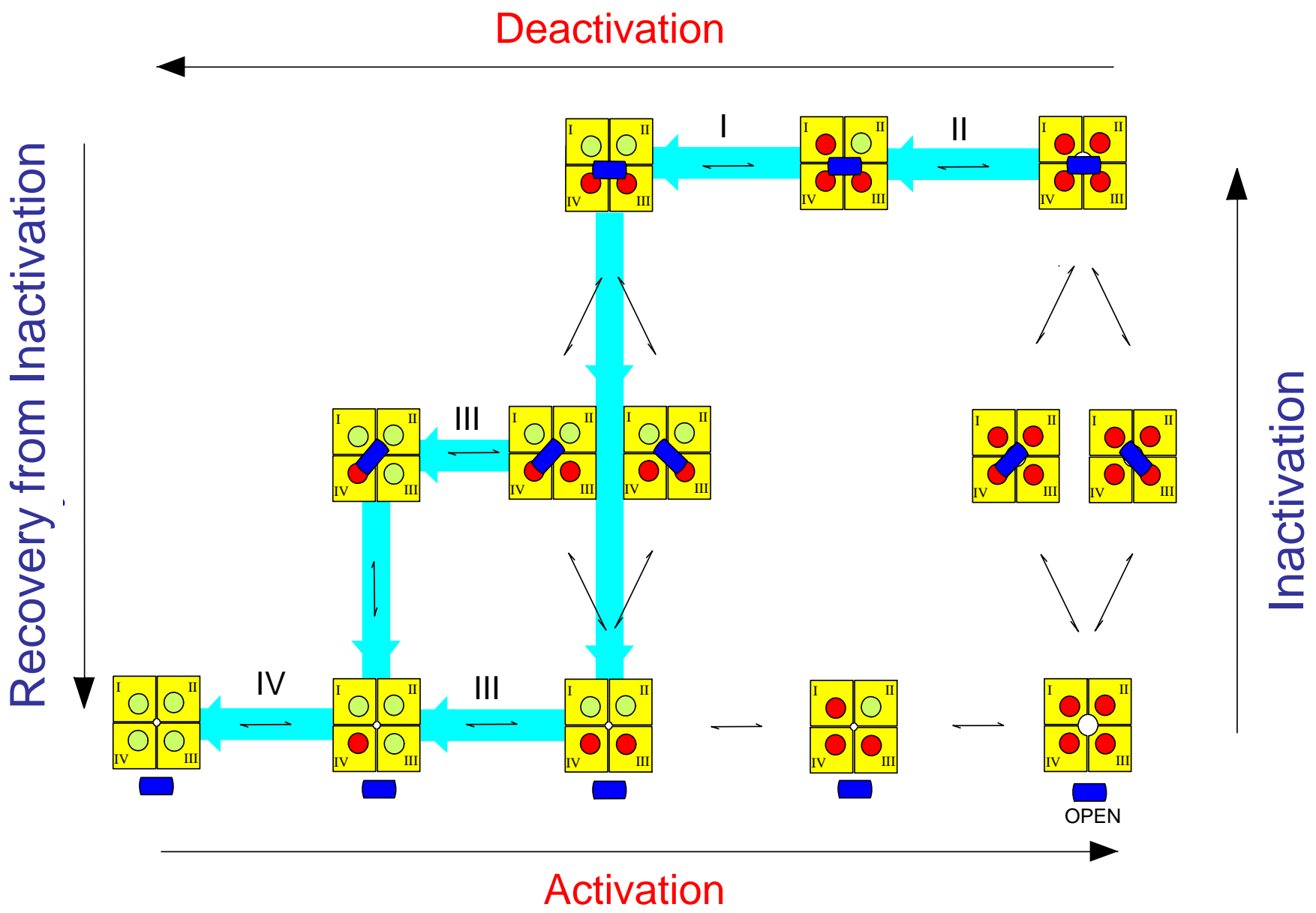


Fast component immobilized by inactivation in Domain IV

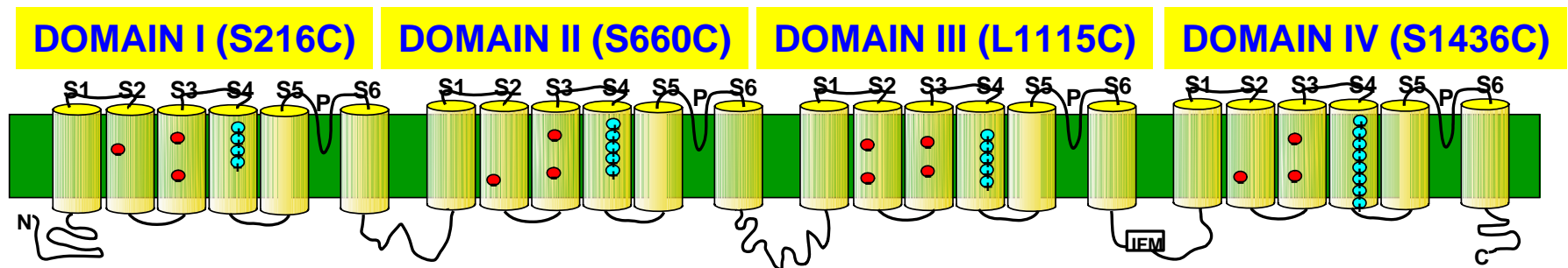
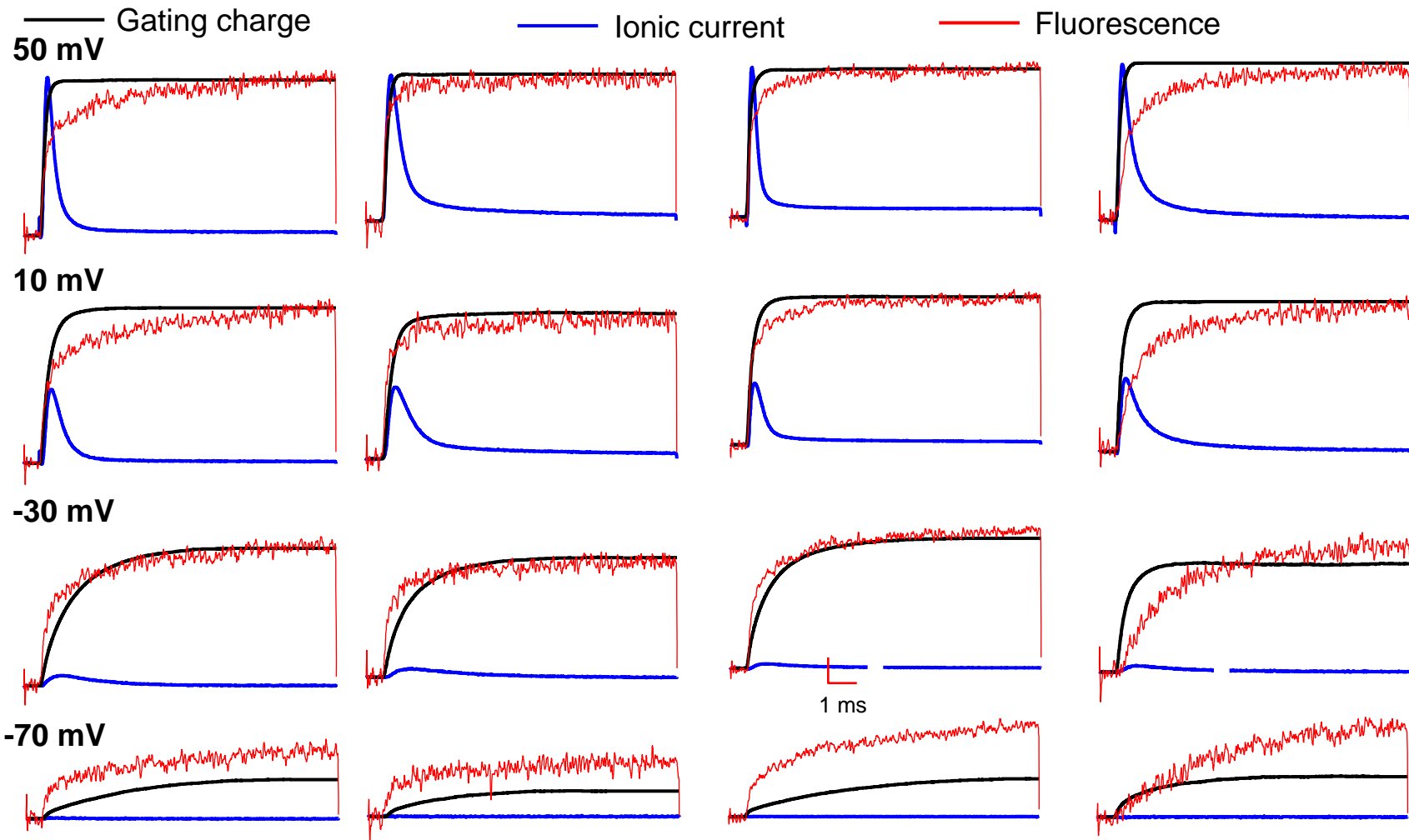


Domain IV (R1448C)

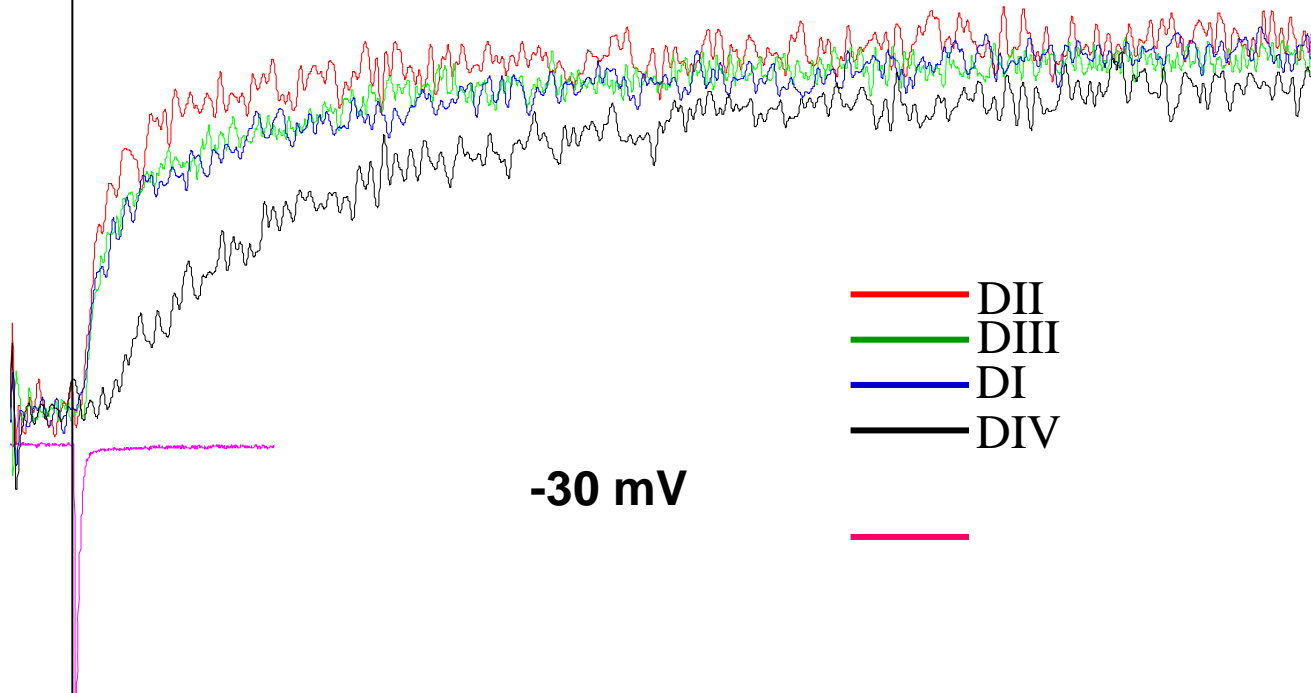




Comparison of fluorescence, gating and ionic traces of labeled mutants

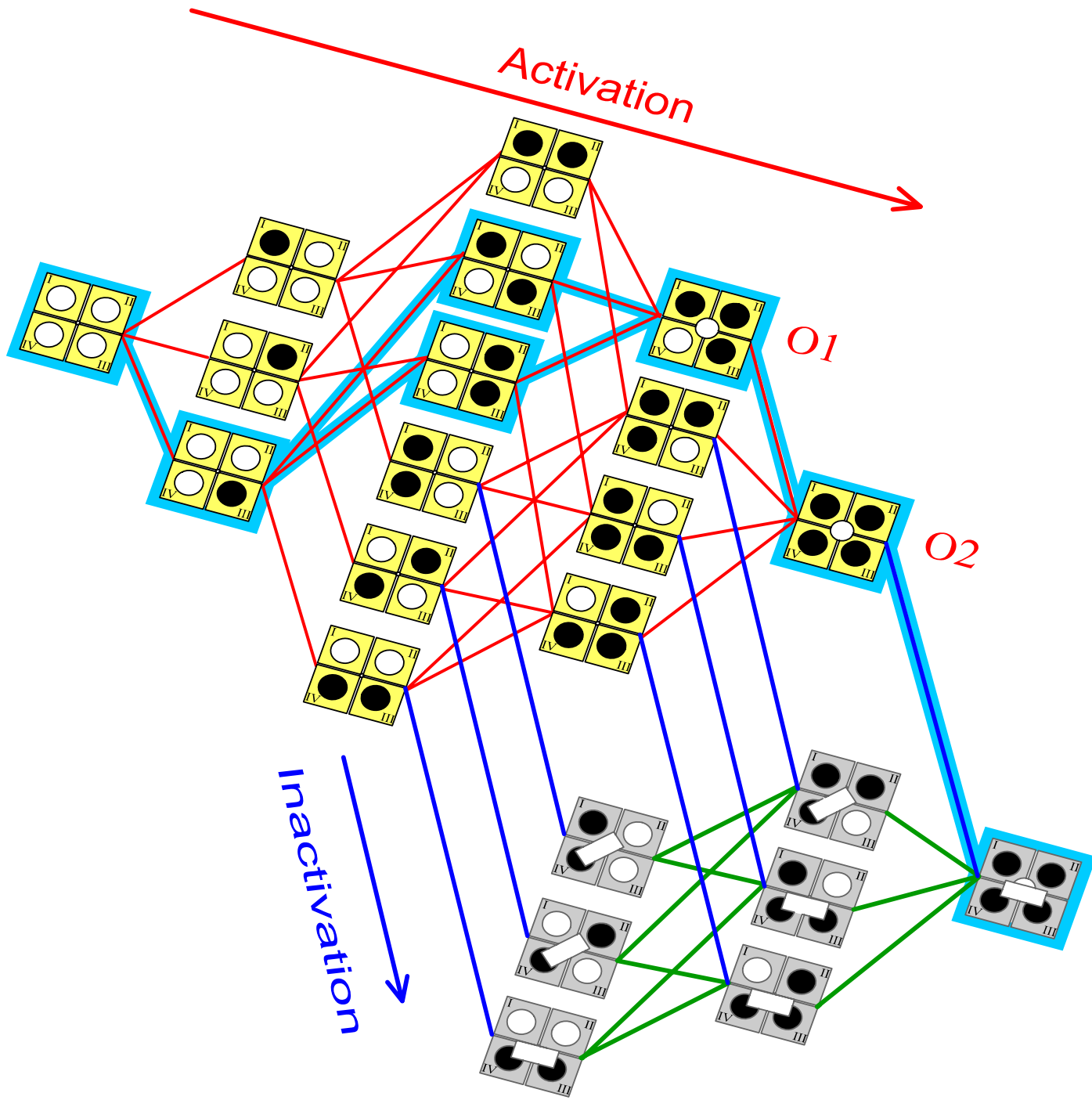


Time course of F signal Of different domains



The S4s of the first three domains do not activate in a specific sequence.

The activation of S4 in domain IV requires a prior activation of one of the three S4 segments.



Activation

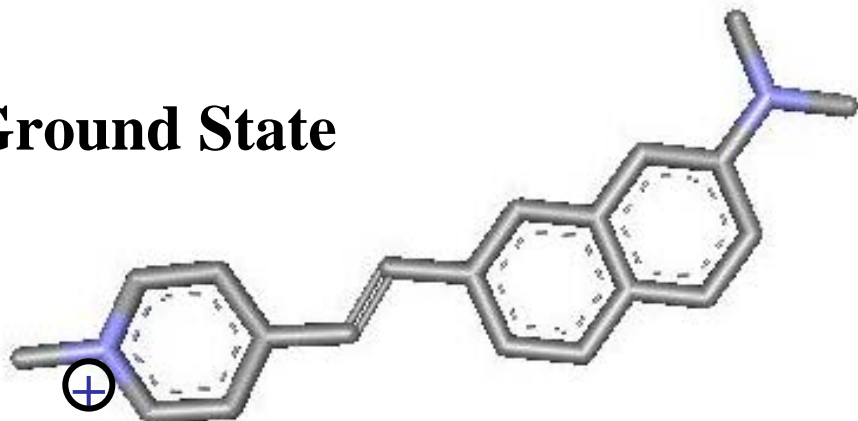
O1

O2

Inactivation

Electrochromic Effect

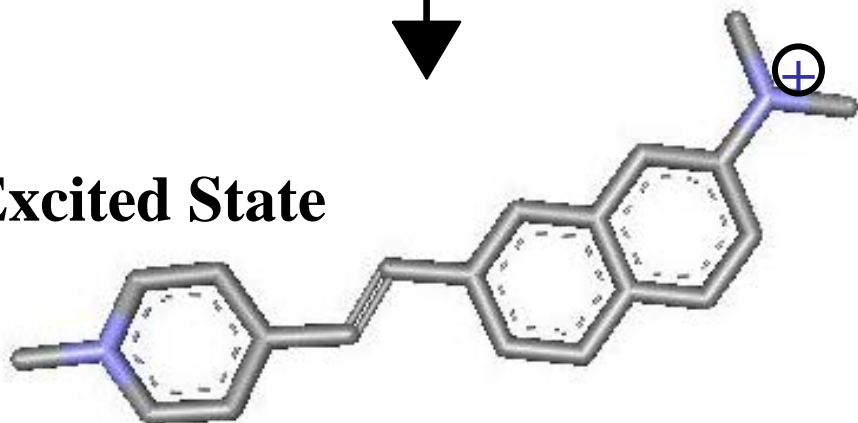
Ground State



Photoexcitation



Excited State



Electrochromic Shift

$$\Delta 1/\lambda = \Delta 1/\lambda^{\text{solv}} - (\mu_e - \mu_g) \mathbf{E} \cos \alpha$$

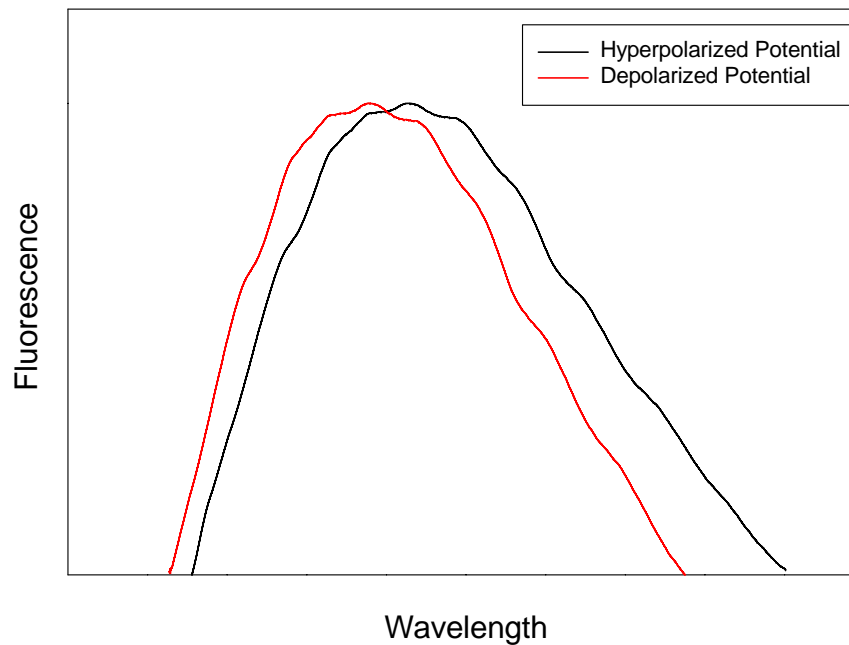
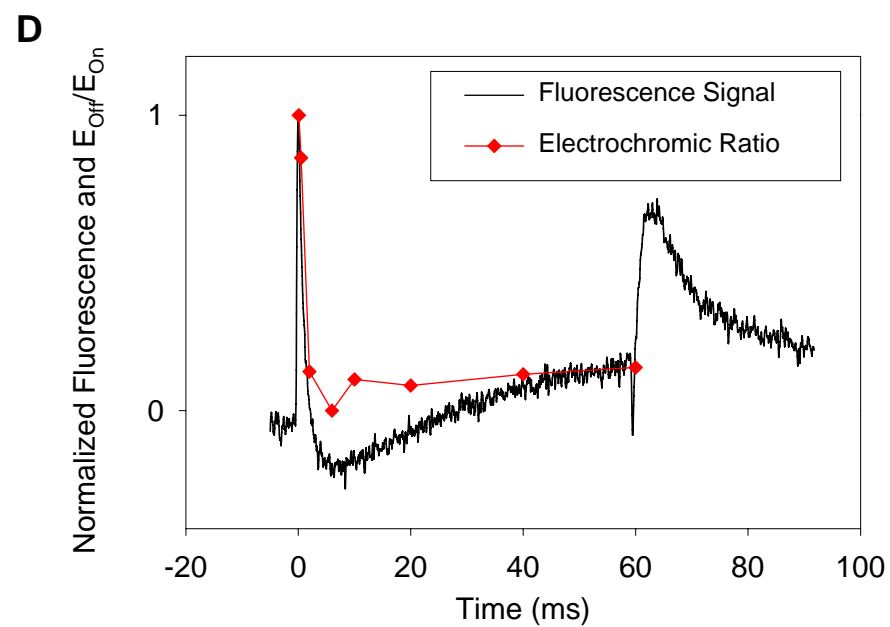
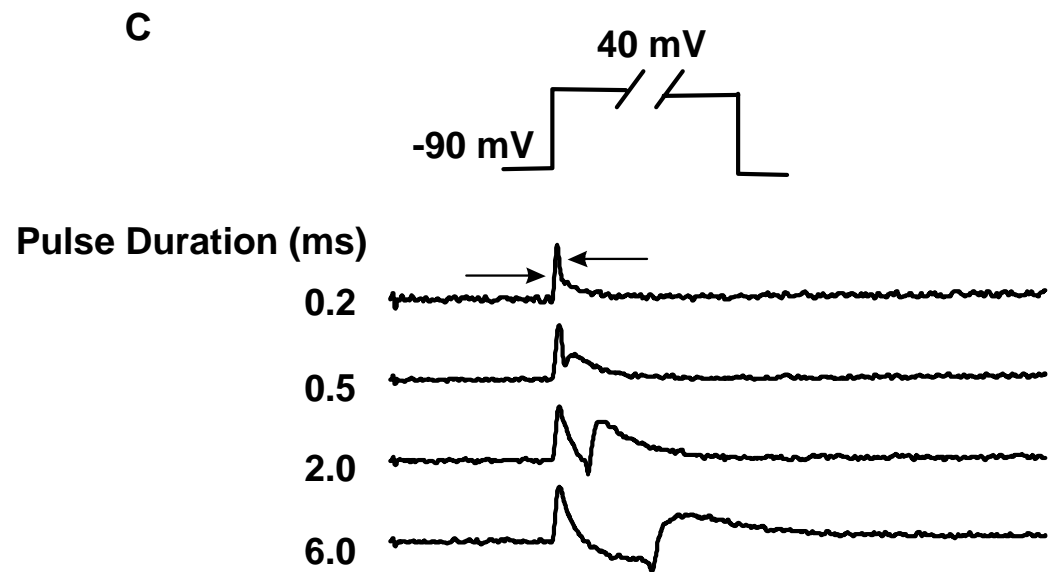
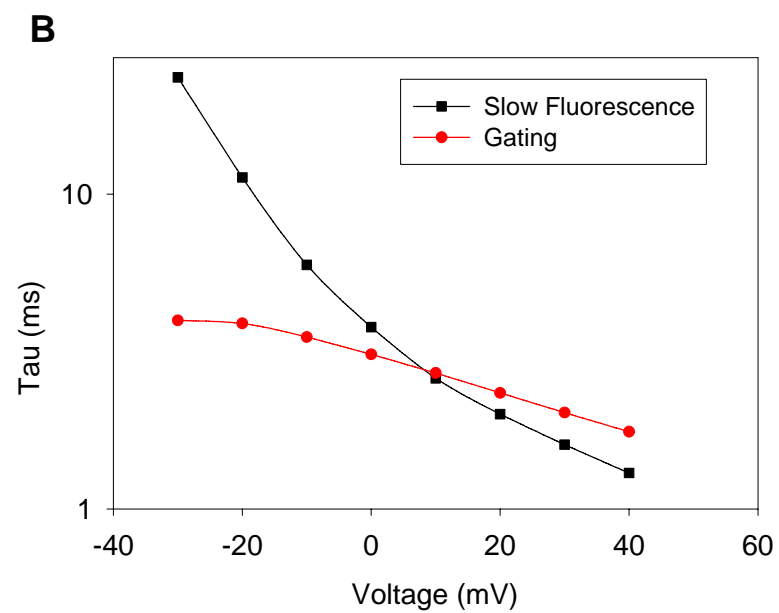
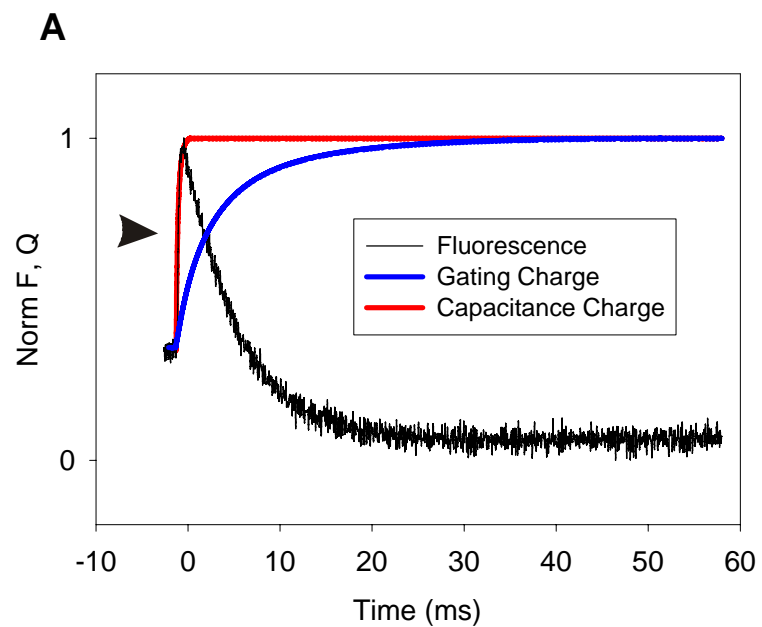
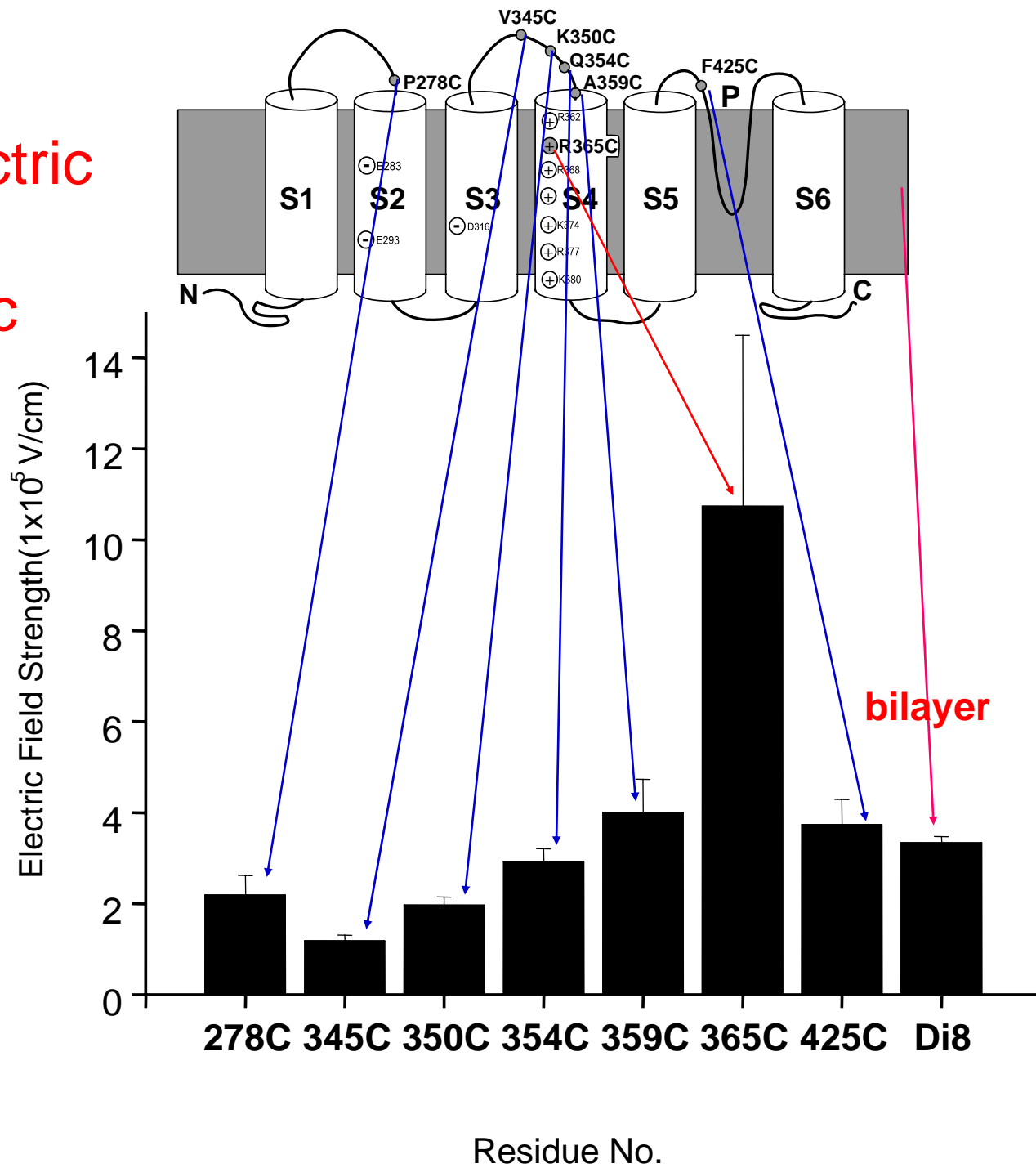


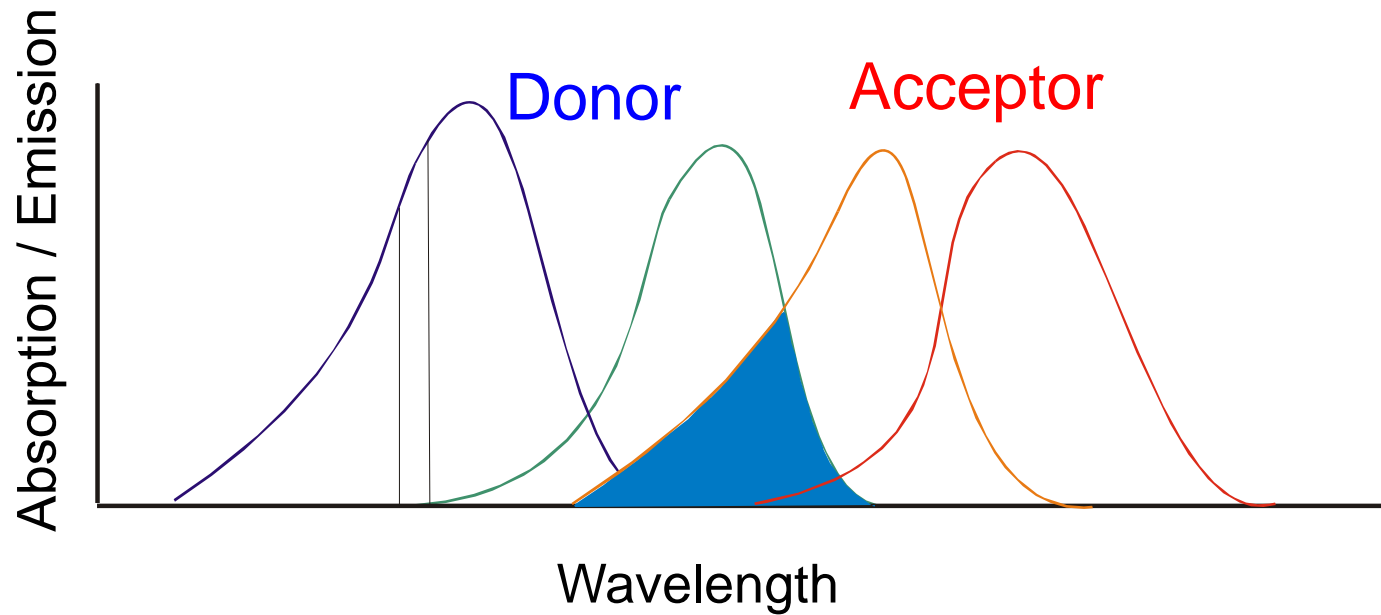
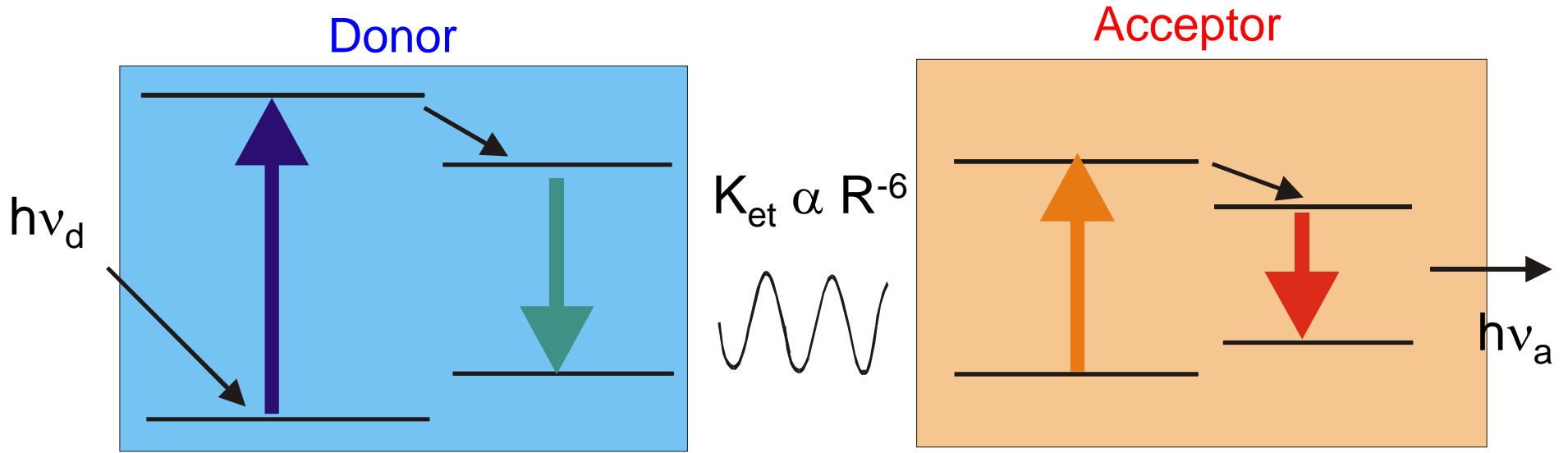
Figure 3



Mapping the electric Field with electrochromic probes



FLUORESCENCE RESONANCE ENERGY TRANSFER (FRET)



$$E = k_{ET} / (k_{ET} + k_n)$$

Forster theory:

$$k_T(R) = \frac{1}{\tau_D} \left[\frac{R_0}{R} \right]^6$$

Where

$$R_0 = 8.79 \times 10^{-5} (J_D q_D n^{-4} \kappa^2)^{1/6}$$

Then,

$$R = R_0 [(1-E)/E]^{1/6}$$

$$R = R_0 [\tau_{DA} / (\tau_D - \tau_{DA})]^{1/6}$$

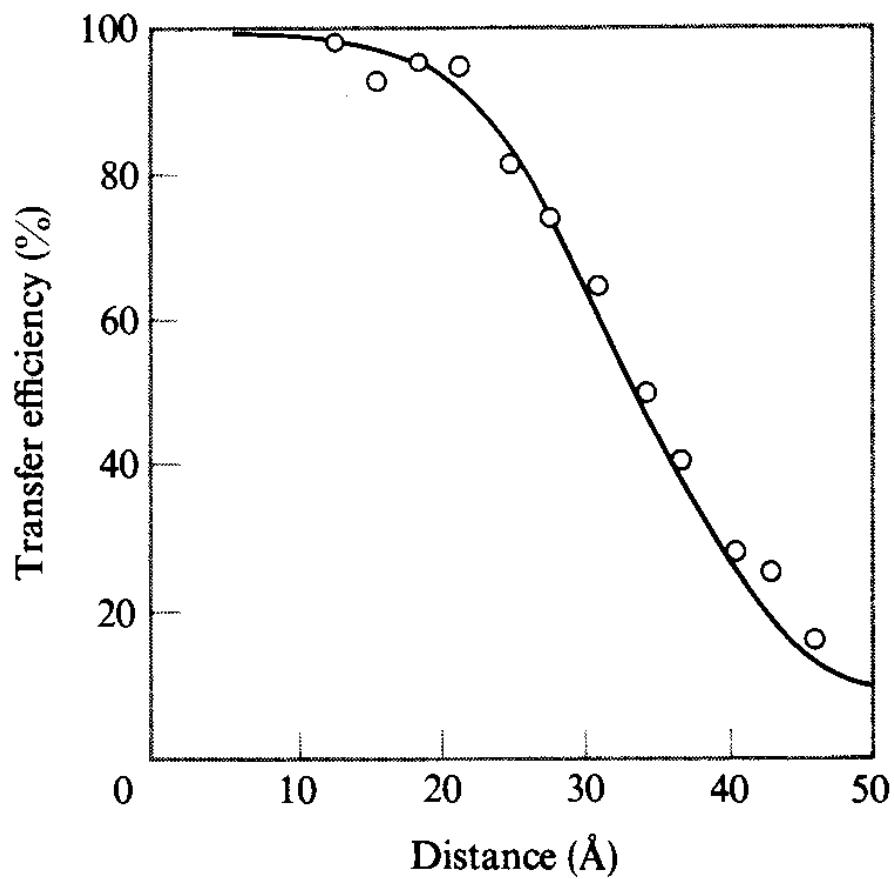
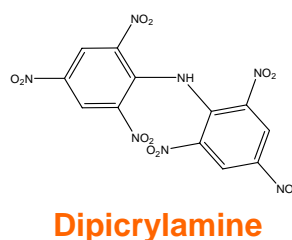
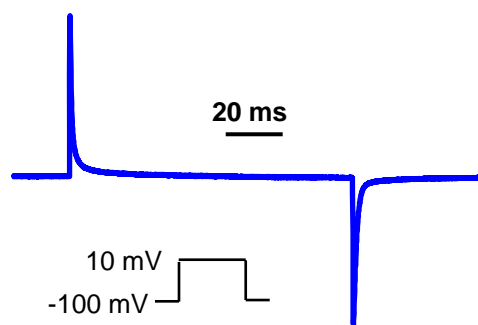


Figure 8-20

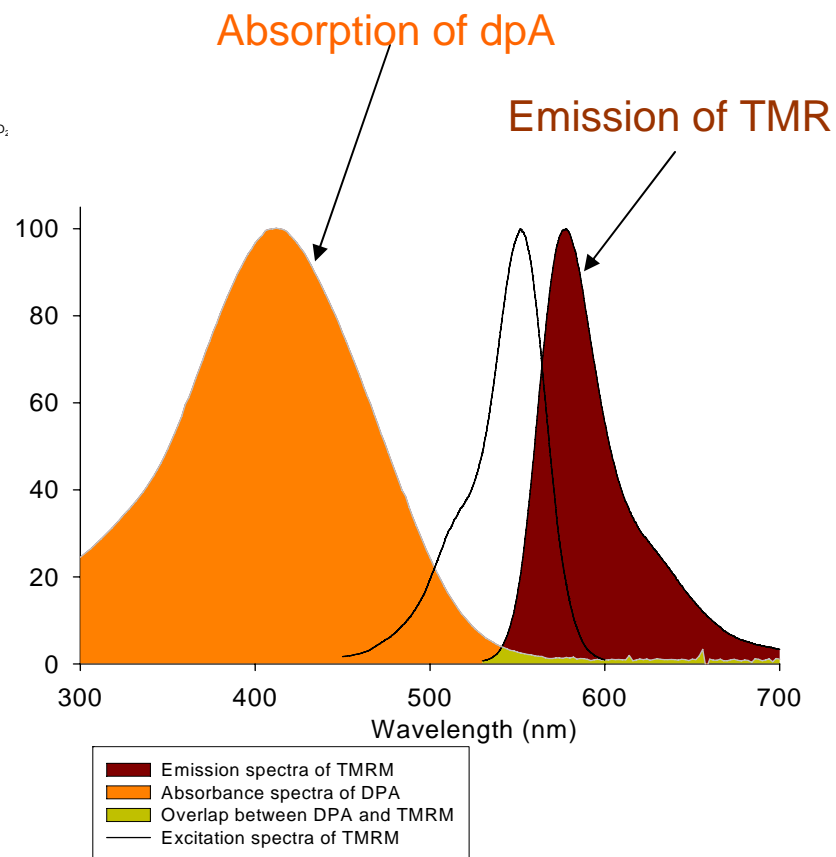
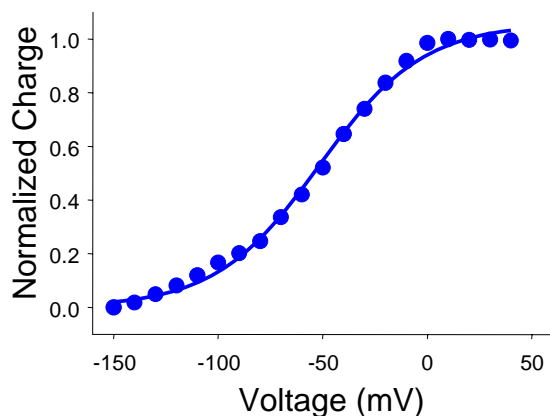
*Efficiency of energy transfer as a function of distance in dansyl-(L-prolyl)_n- α -naphthyl semicarbazide oligomers with $n = 1$ to 12. The curve was fit to the data with Equation 8-57. [From L. Stryer and R. P. Haugland, *Proc. Natl. Acad. Sci. USA* 98:719 (1967).]*

Another method: Using hydrophobic ions as FRET acceptors (reference) in the bilayer

Gating current trace of
Dipicrylamine (dpA)

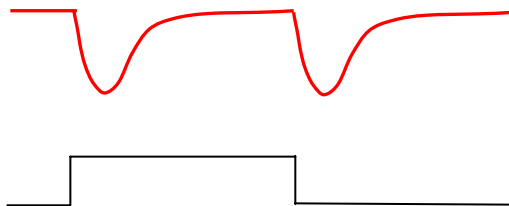
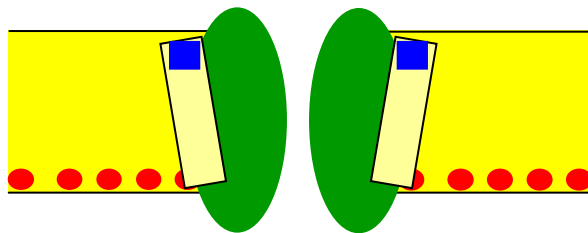
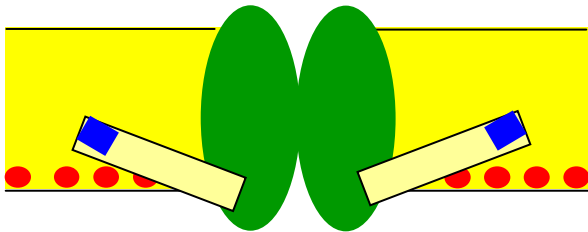
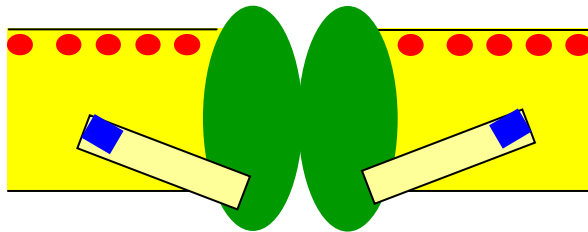


Charge voltage relationship of dpA in
oocyte membranes

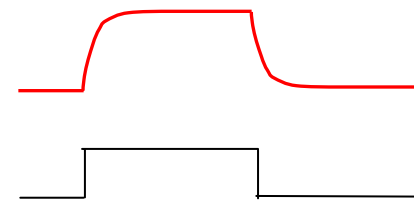
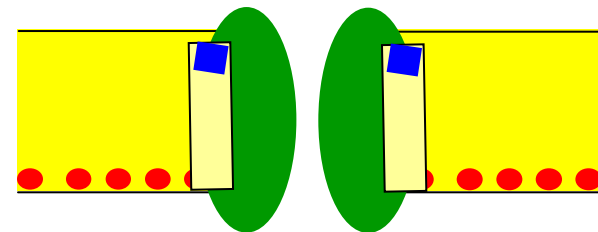
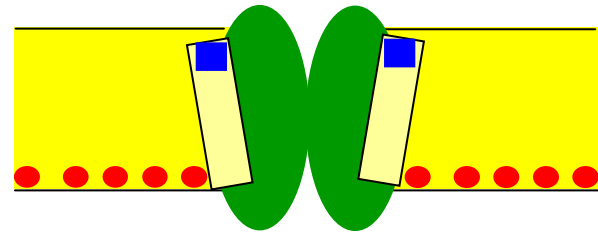
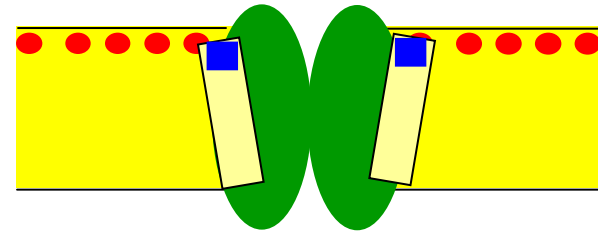


Calculated R_0 between
TMR and dpA is 20 Å

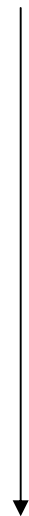
Large movement



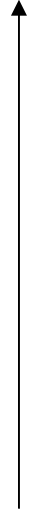
Small movement



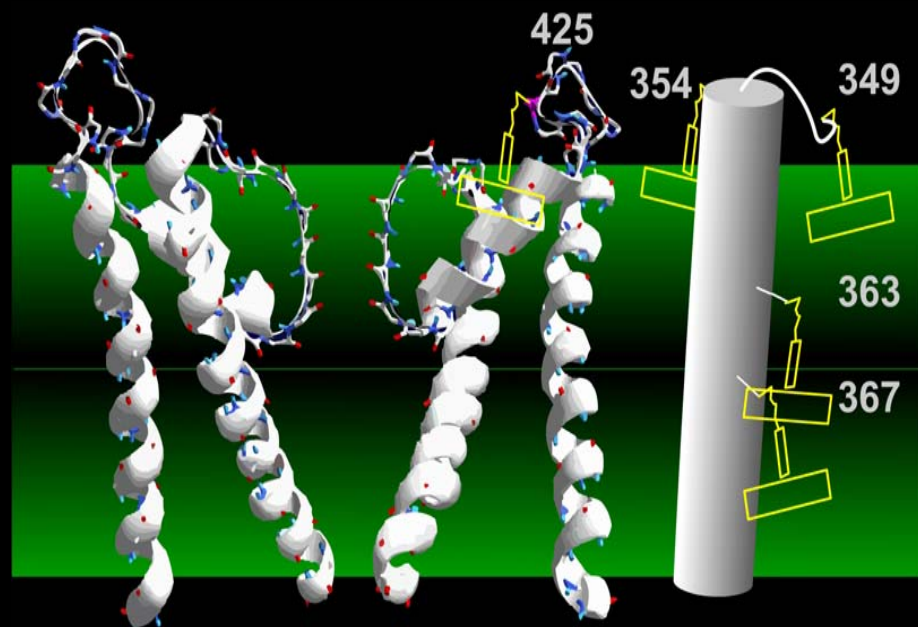
depolarization



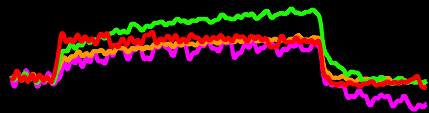
repolarization



average



363C - ABD



1% $\Delta F/F$

20 ms

4% $\Delta F/F$

20 ms

WT

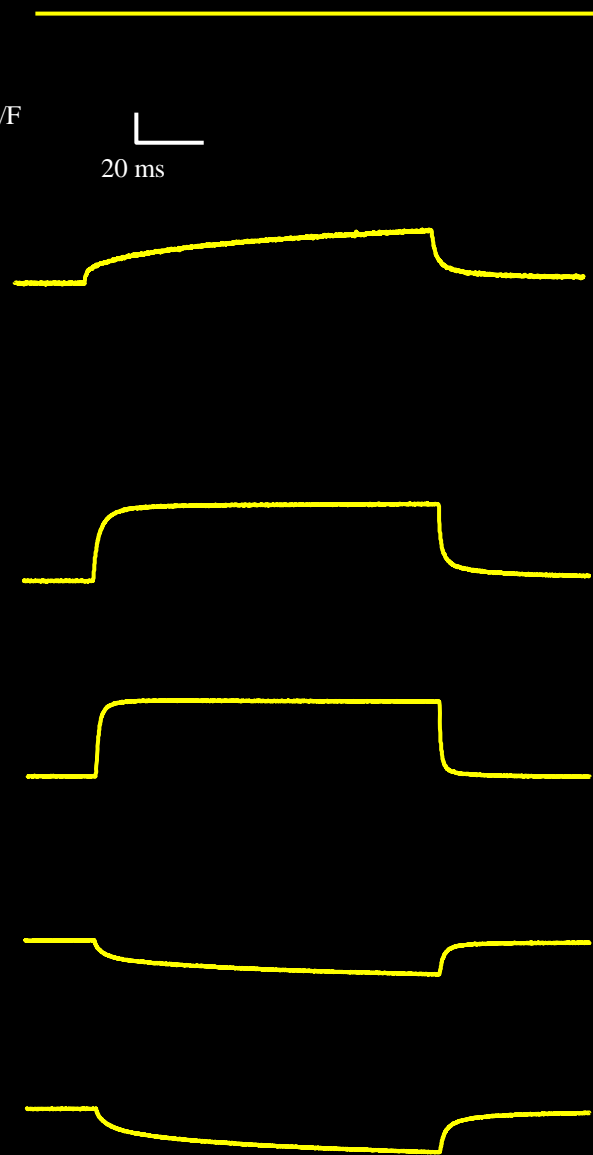
425C

349C

354C

363C

367C



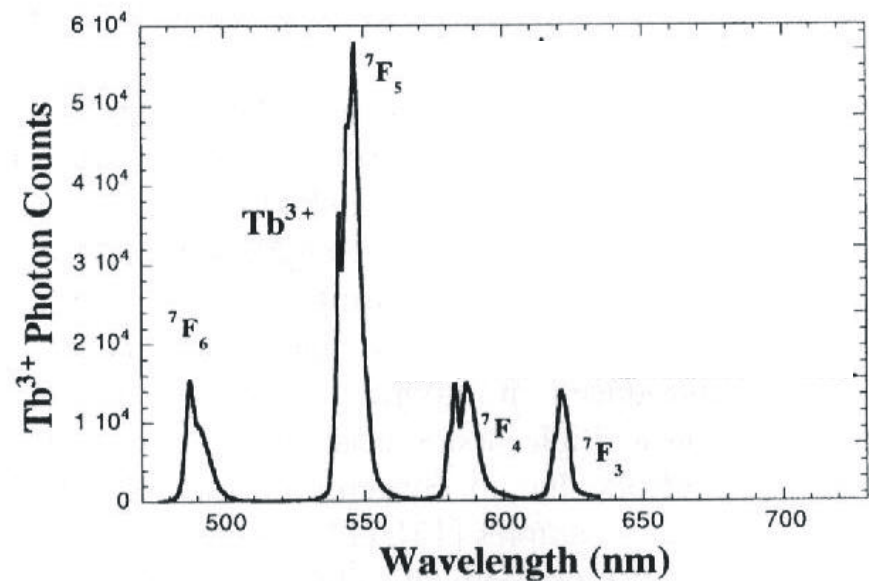
Lanthanide-based Resonance Energy Transfer (LRET)

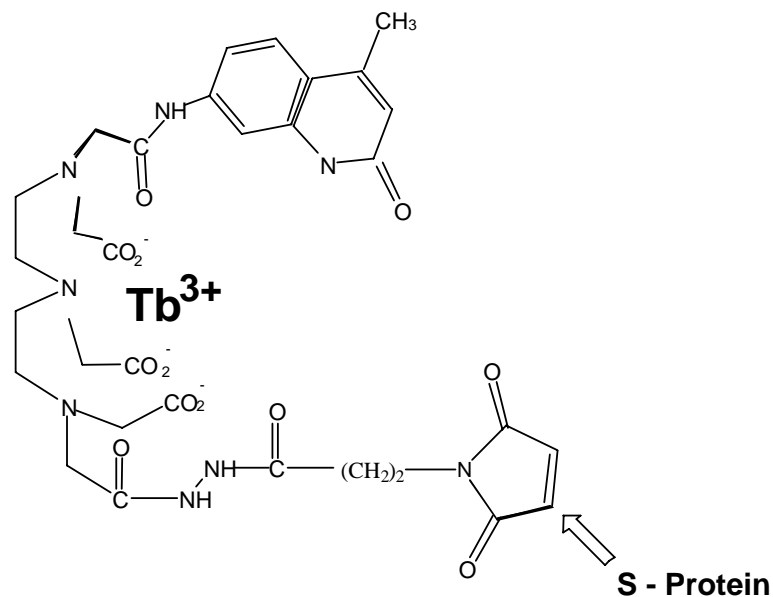
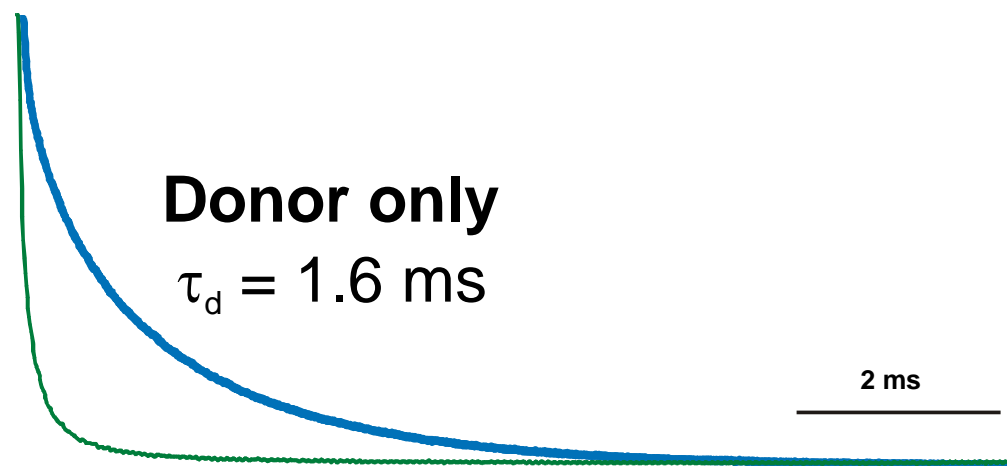
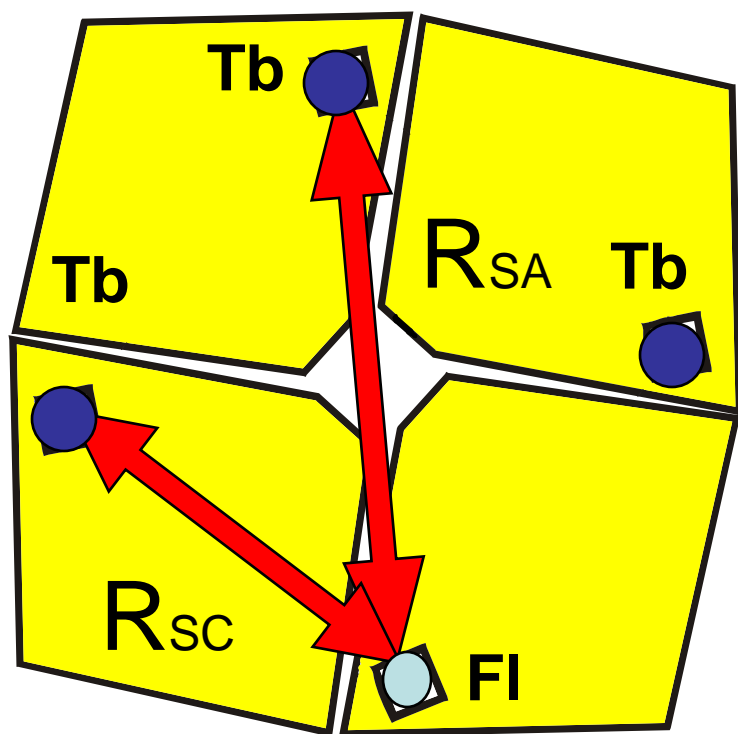
LRET utilizes luminescent lanthanides (i.e. Tb^{3+} , Eu^{3+}) with sensitizer and reaction group as fluorescent donors

- Lifetimes are easily measurable ($\tau \sim 0.2 - 1.5$ ms)

- Lanthanides fluoresce isotropically, reducing errors from orientation factor κ^2

- Sharply spiked emission spectrum and temporal elimination of direct acceptor excitation enables specific measurements of acceptor sensitized emission





Sensitized emission

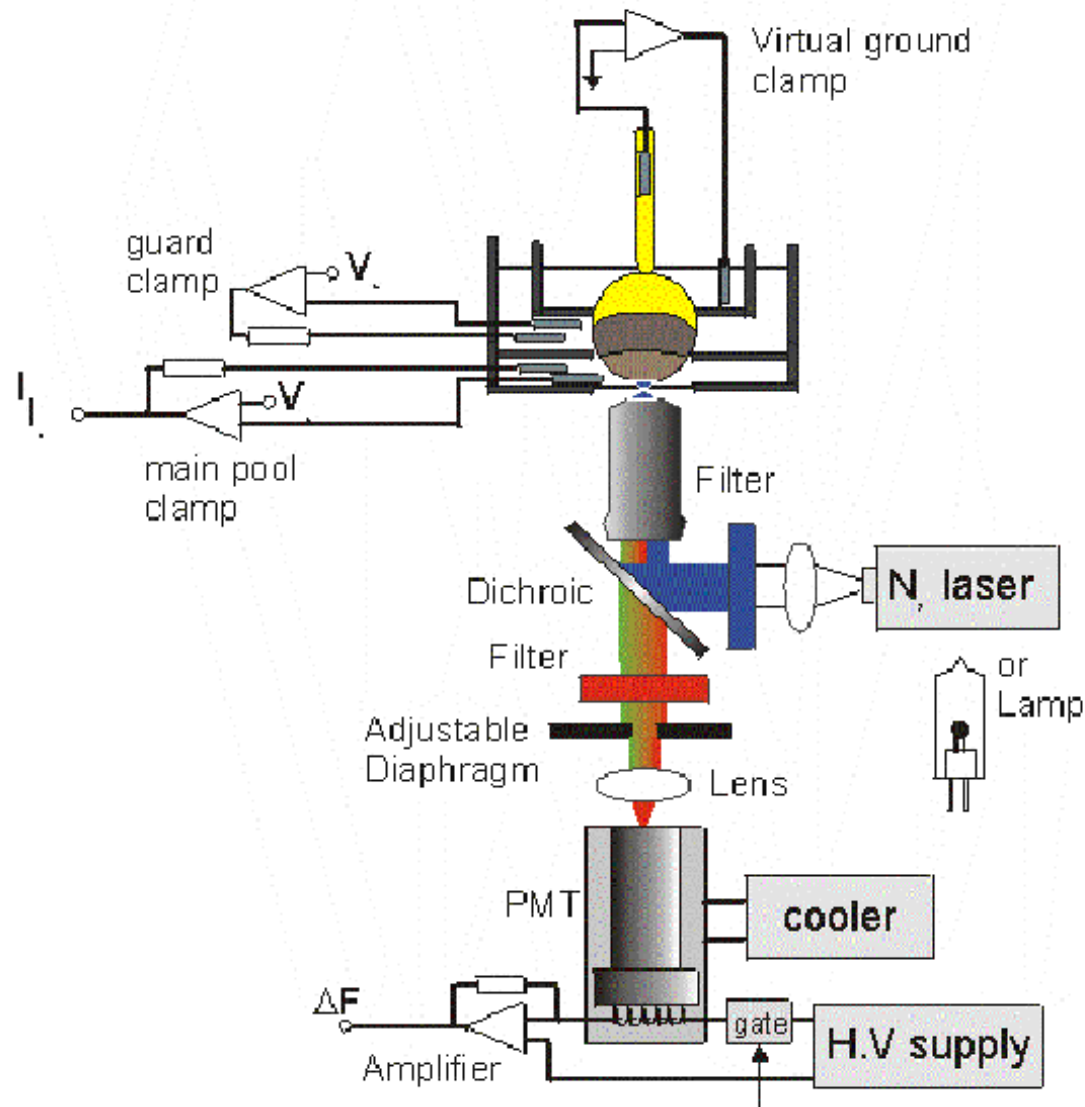
$$\tau_{ad} = 92 \mu\text{s}, 566 \mu\text{s}$$

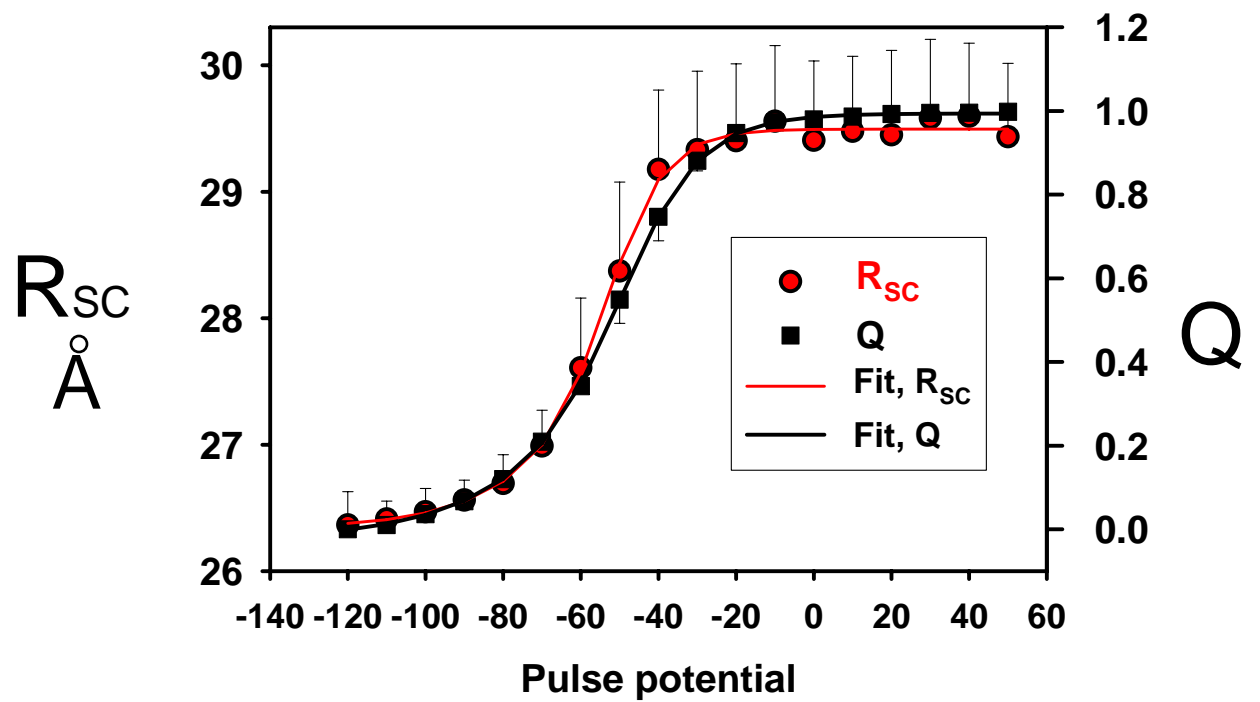
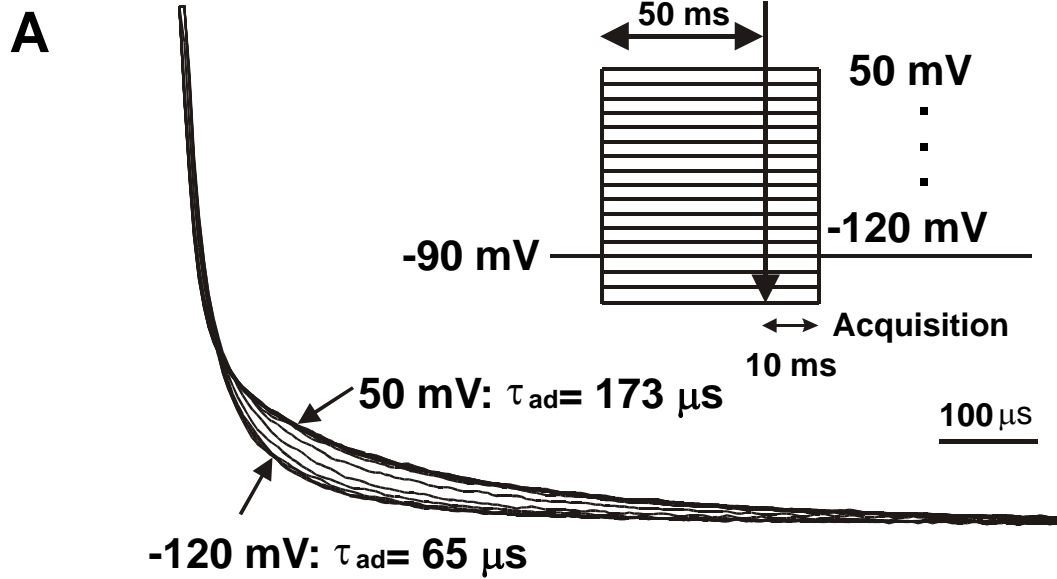
$$R_{SC} = 28 \text{ \AA}$$

$$R_{SA} = 41 \text{ \AA}$$

R_{SA} predicted from R_{SC} using
 Pithagorean Theorem = 40 A

INVERTED CUT-OPEN FLUORESCENCE SETUP





The results show that upon depolarization distances from toxin to:

S4 decrease by 0.8 Å

(361 and 365 in S4)

S3-S4 linker decrease by 2 Å

(351, 352, 353)

S3 Increase by 1 Å

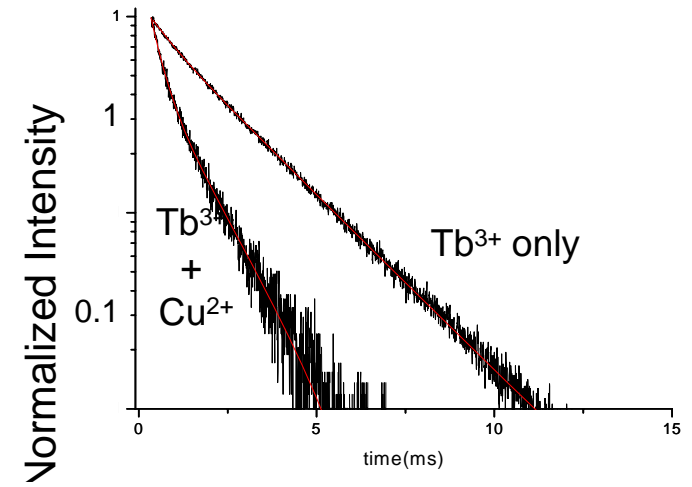
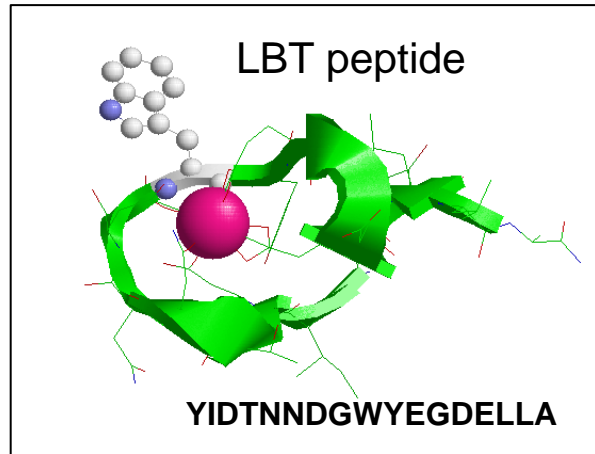
(333, 335)

Conclusion:

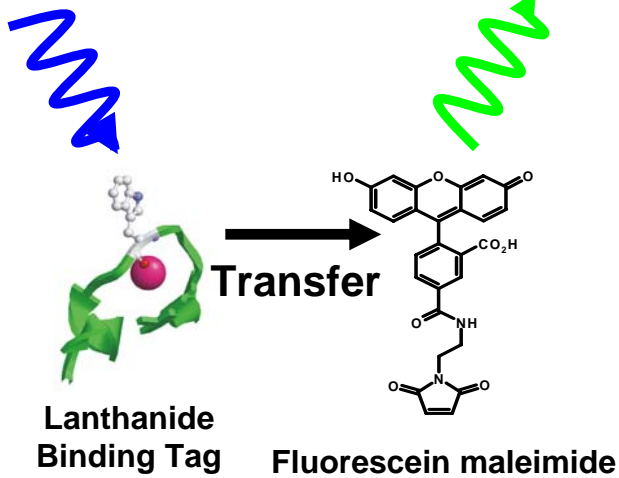
transmembrane displacement is about 2 Å

With S4 and S3 moving in opposite directions

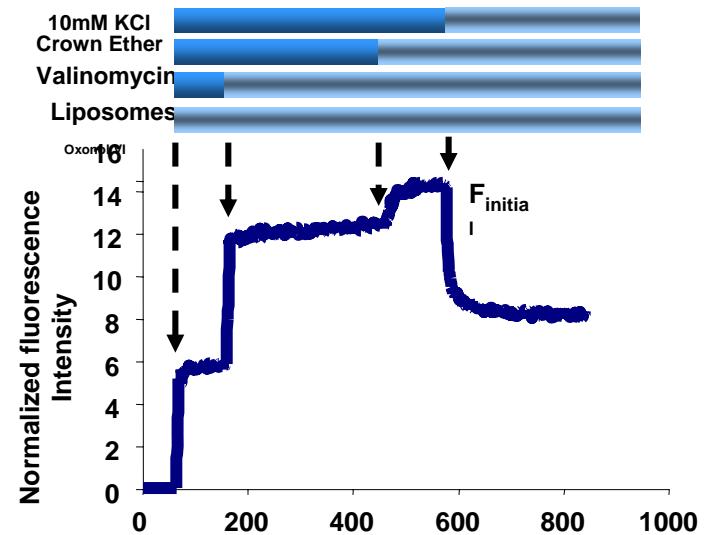
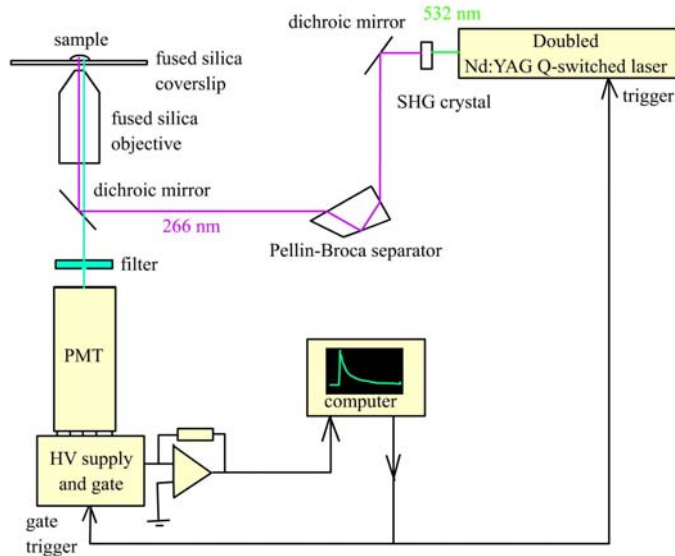
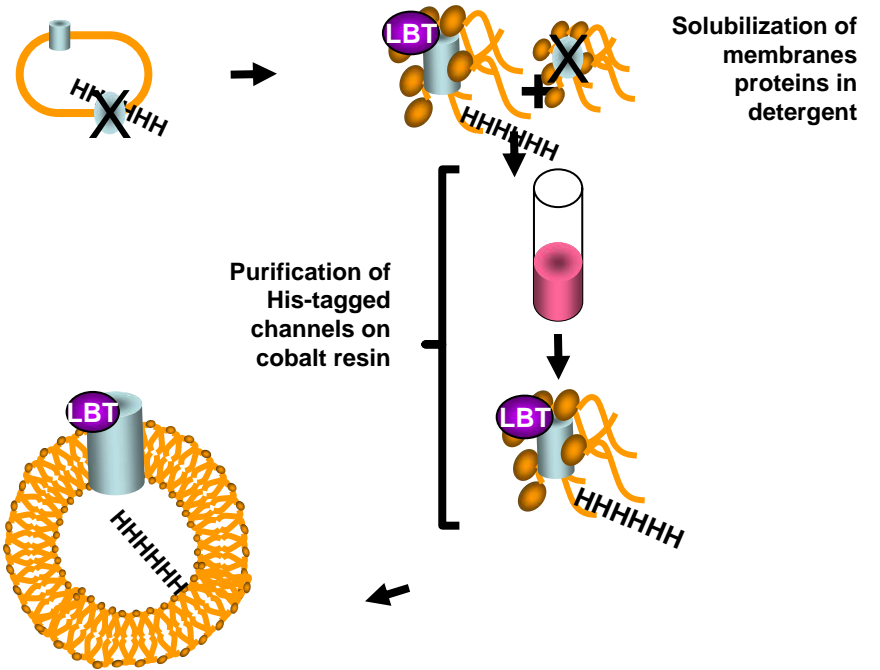
Genetically encoded LRET



Lanthanide based Transfer-LRET using a genetically encoded tag



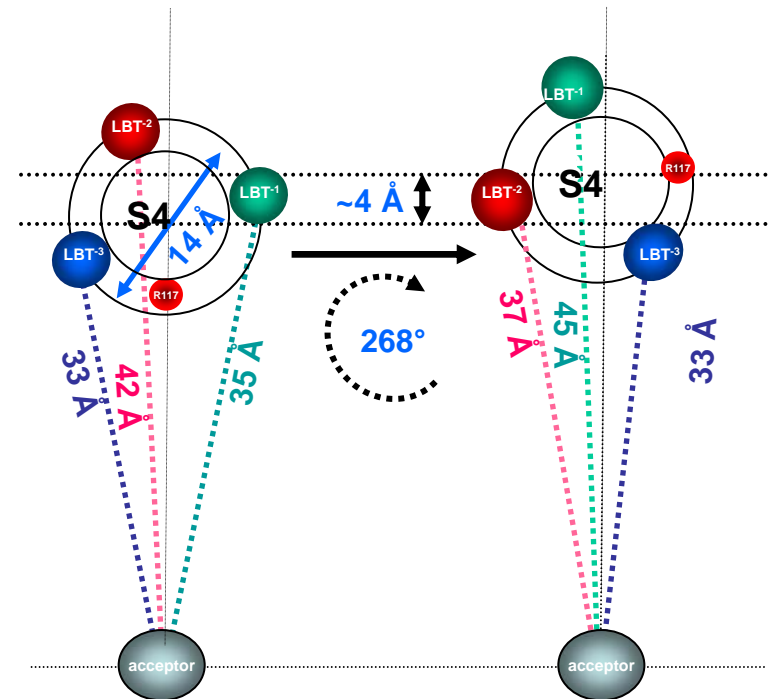
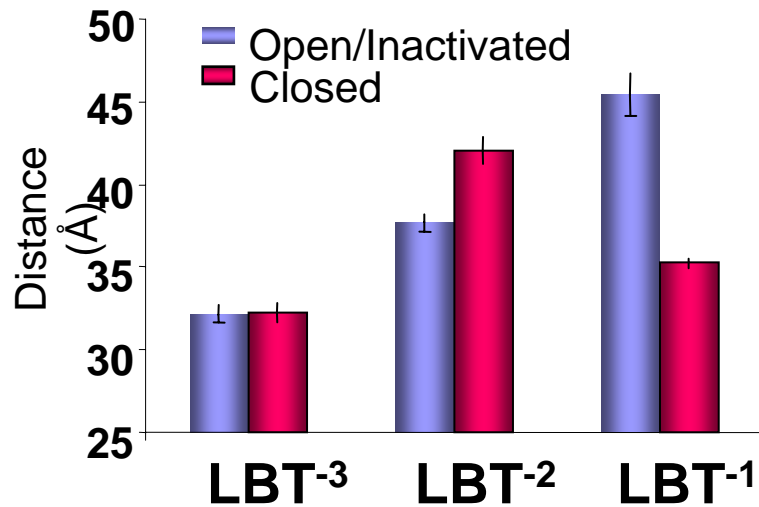
Expression of modified channels in XL-1 Blue



Example:

Demonstration of rotational movement of the voltage sensor of KvAP

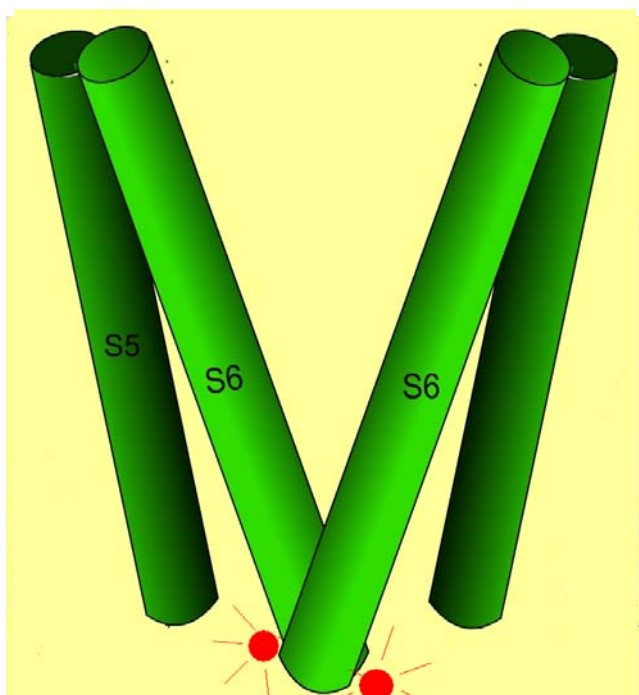
Distance changes in KvAP



These experiments are also done in oocytes under voltage clamp allowing better potential control and time resolution

Conformational changes detected at the single molecule level

Optical Methods: fluorescence



SPR & SPCE

- Surface plasmon resonance (SPR) is used to excite fluorophores near a specially prepared thin metal surface on glass substrate
- Surface plasmon-coupled emission (SPCE) is used in the configuration below to capture the emission of the fluorophores excited by SPR as described by Lakowicz et al

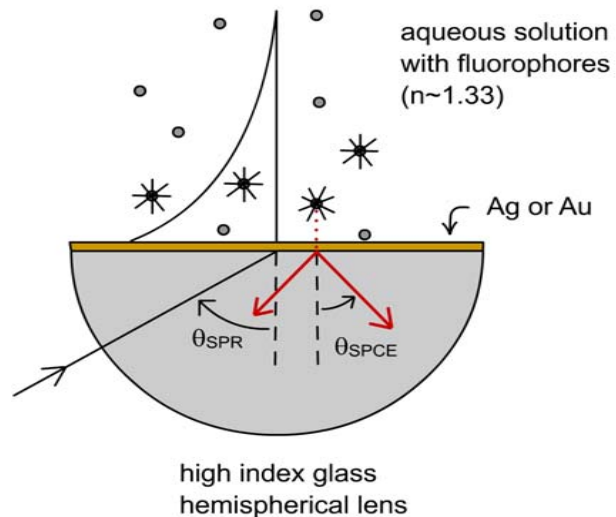


Figure 1. Schematic of the application of surface plasmon resonance to fluorescence in aqueous solution. The evanescent field strength decreases away from the metal surface. Molecules within the evanescent field are excited (stars). However, molecules within approximately 10 nm of the metal surface will be quenched and those beyond the evanescent field are not excited.

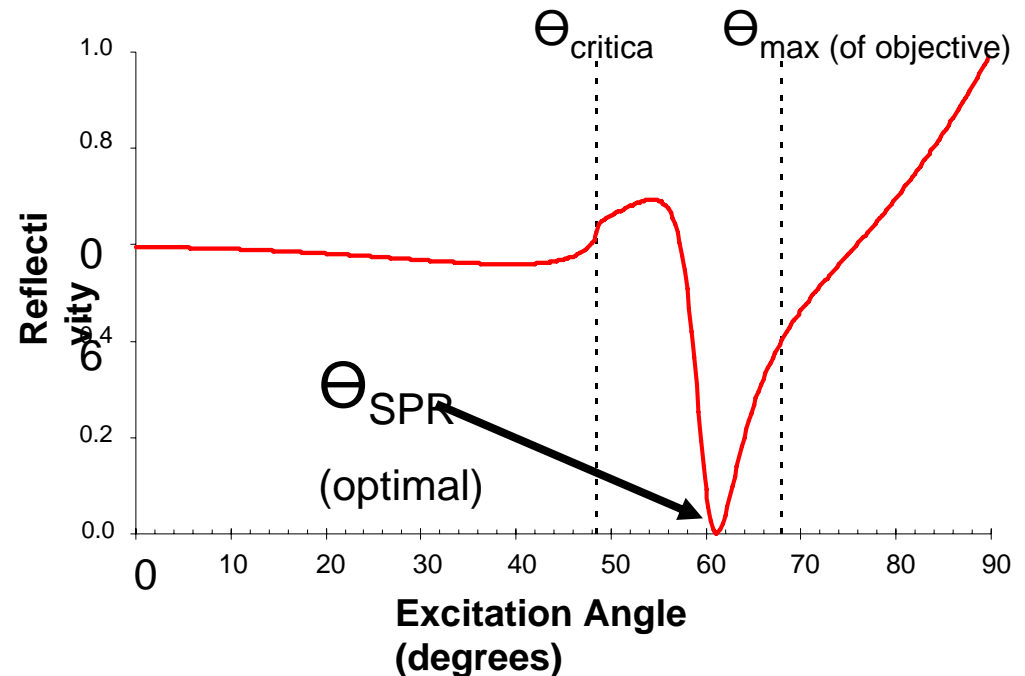
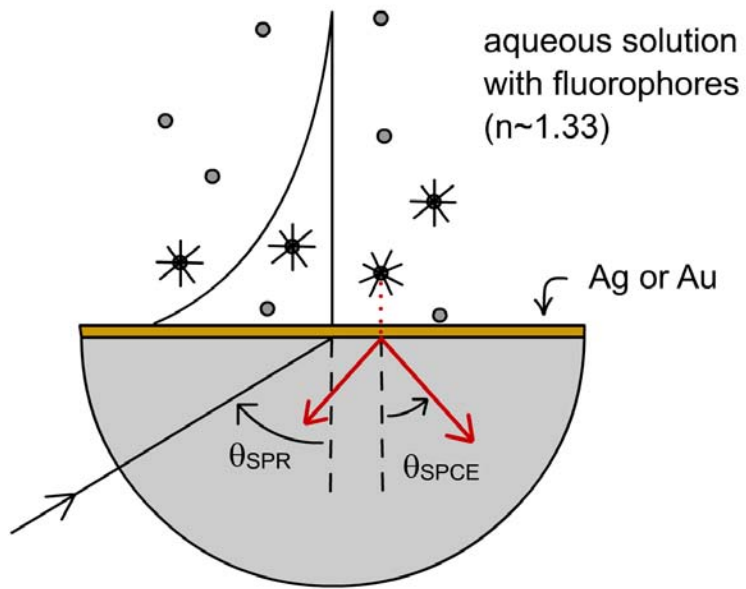
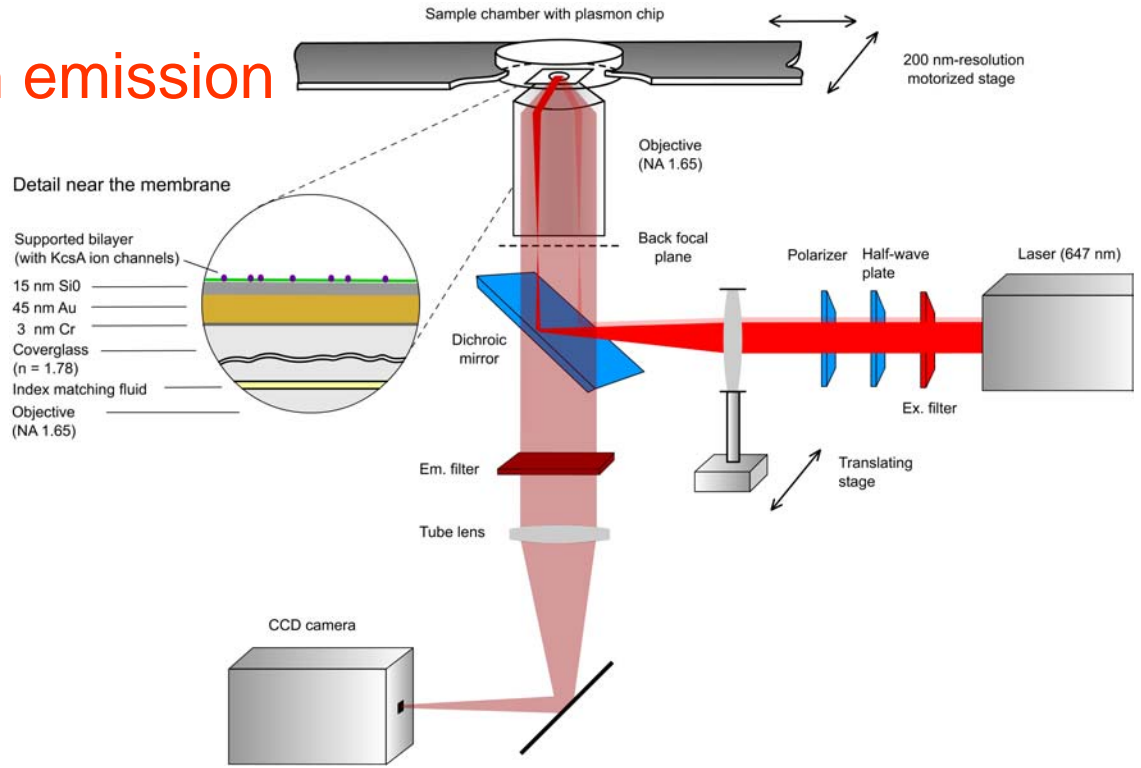


Figure 2. Simulation of reflectance of 647 nm laser excitation off the surface of the plasmon chip. The “dip” is the angle at which photons are most efficiently absorbed and maximally excite surface plasmon. The critical angle for SF11 glass at an H₂O interface and the maximum acceptance angle for the Olympus APO 100x NA 1.65 objective are shown as dotted lines.

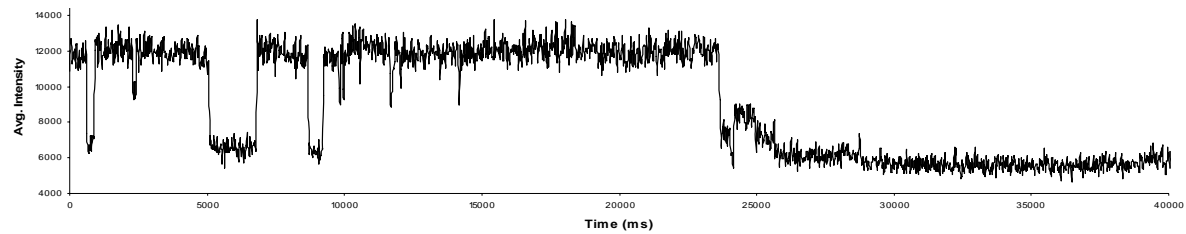
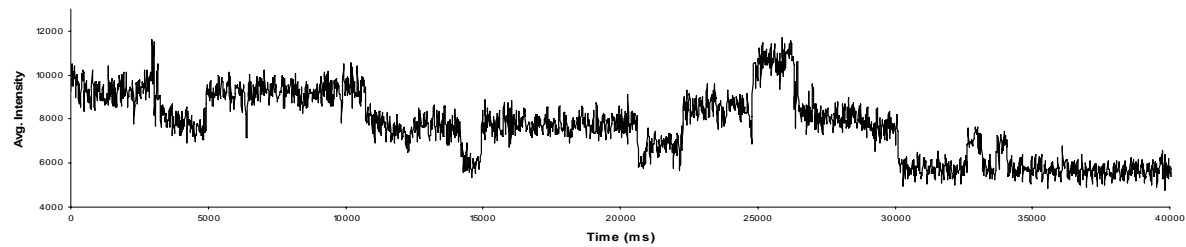
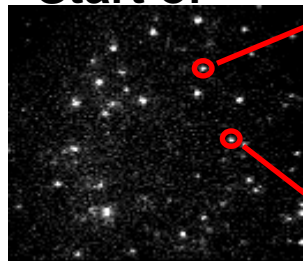
Surface-coupled plasmon emission



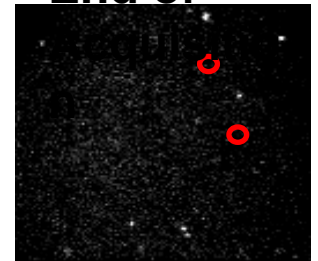
high index glass hemispherical lens



Start of

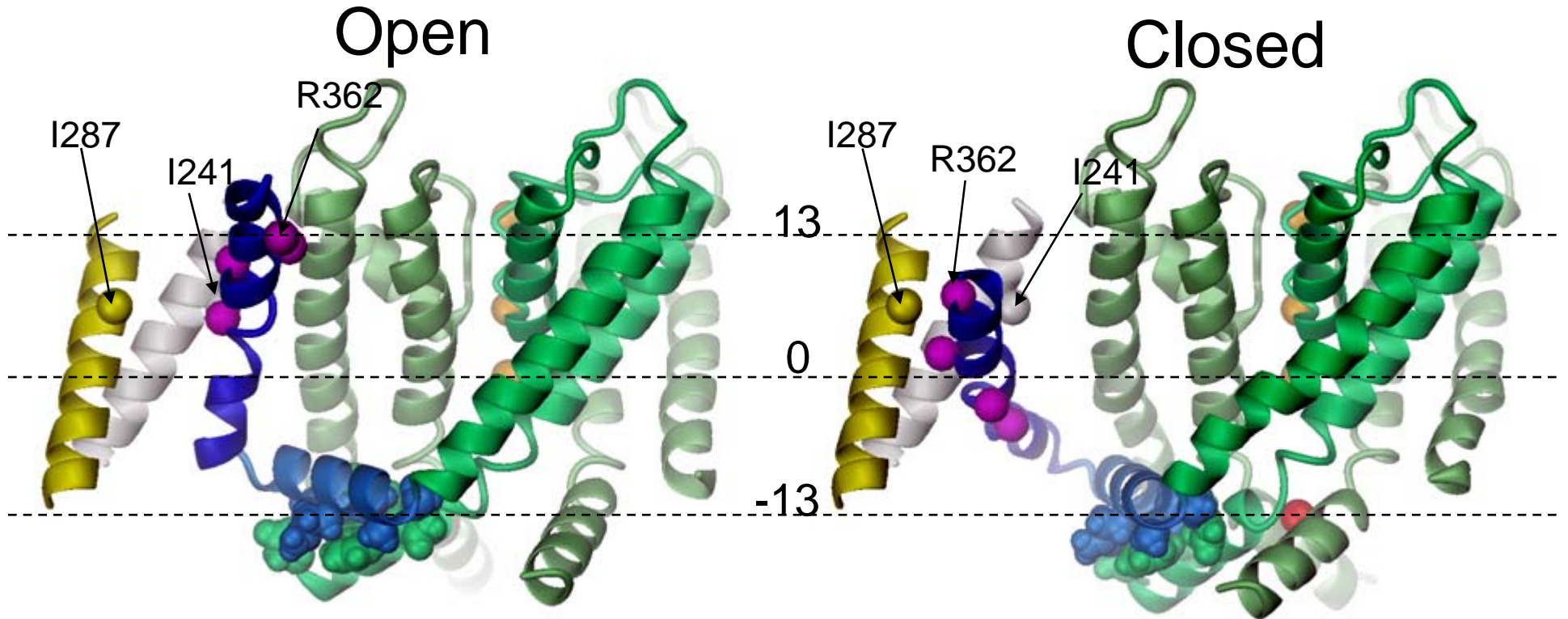


End of



CONCLUSION

The position of S4 at the level of R362 is constrained with respect to S1 and S2 in CLOSED state

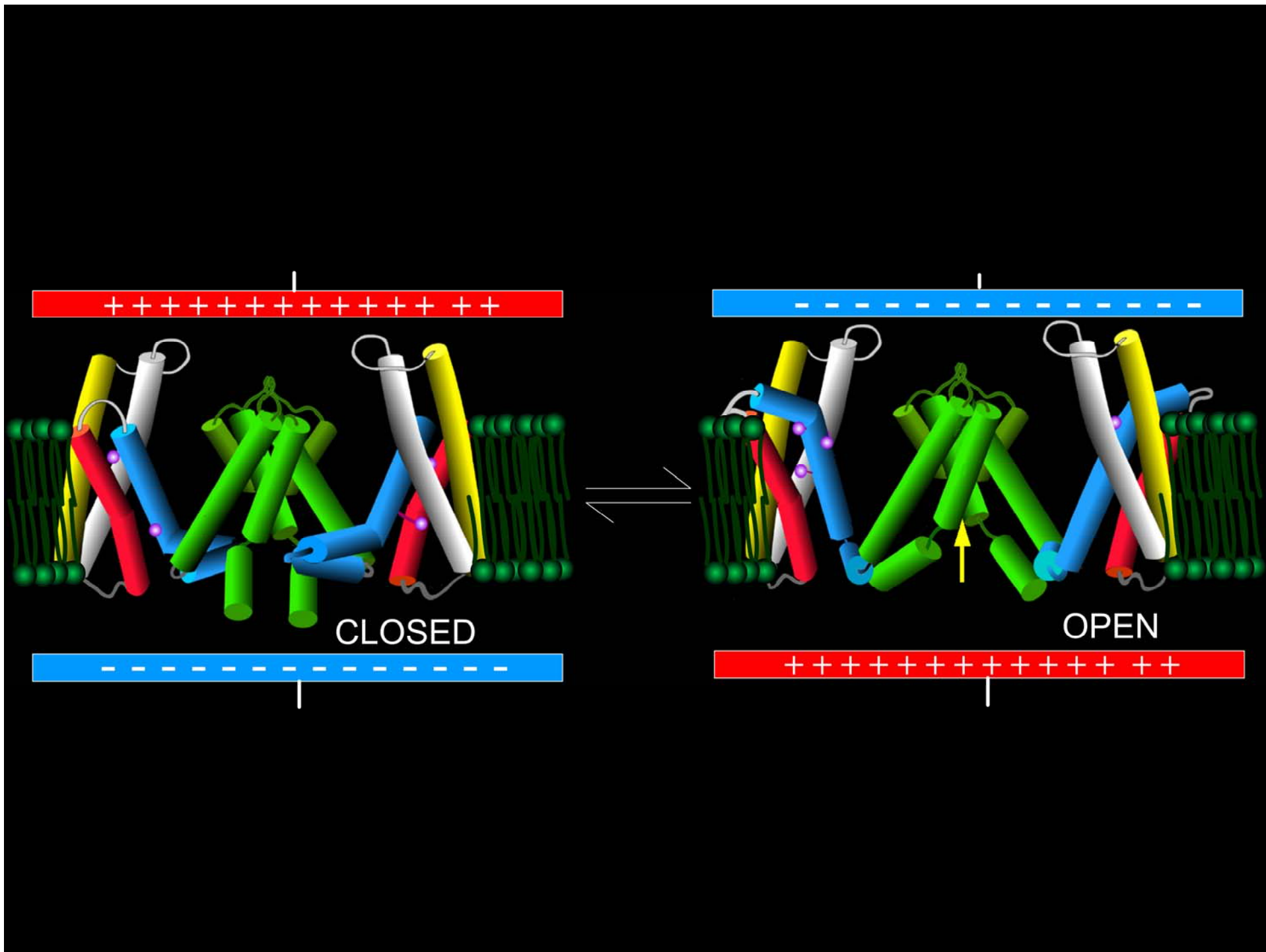


Translation ~6.5 Å

Rotation ~180°

Tilt ~30°

MINIMAL MODEL





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Jerome Lacroix

Fabiana Campos

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Baron Chanda

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